

The Prebiotic Potential and Molecular Diversity of Space Environments

Past Projects and Future Possibilities with SOFIA

Duncan V. Mifsud^{1,2} and Sergio Ioppolo³

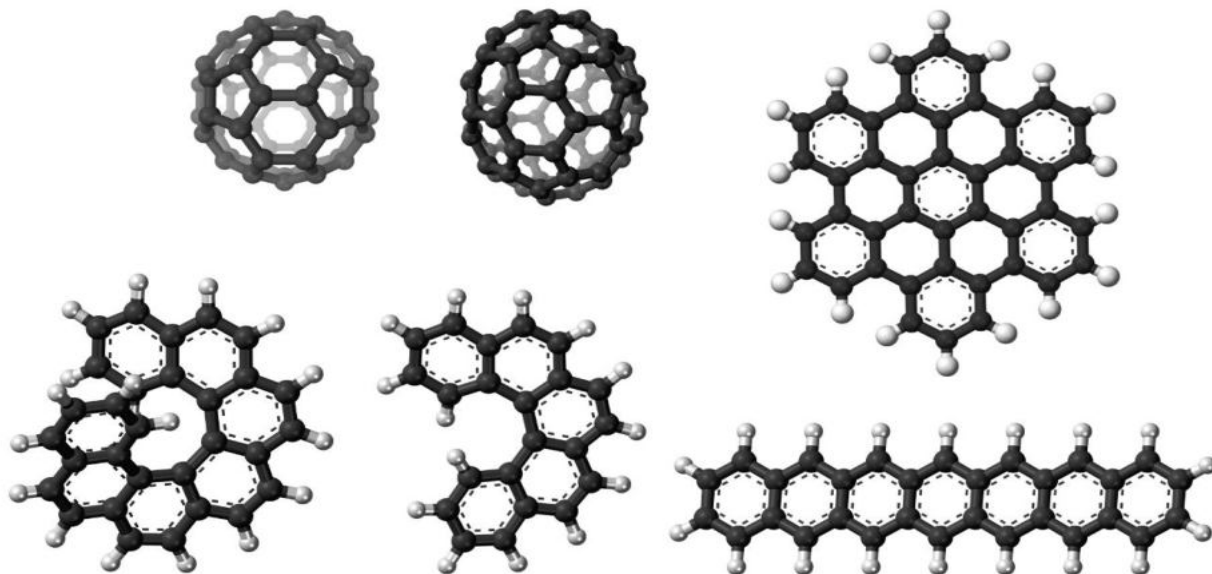
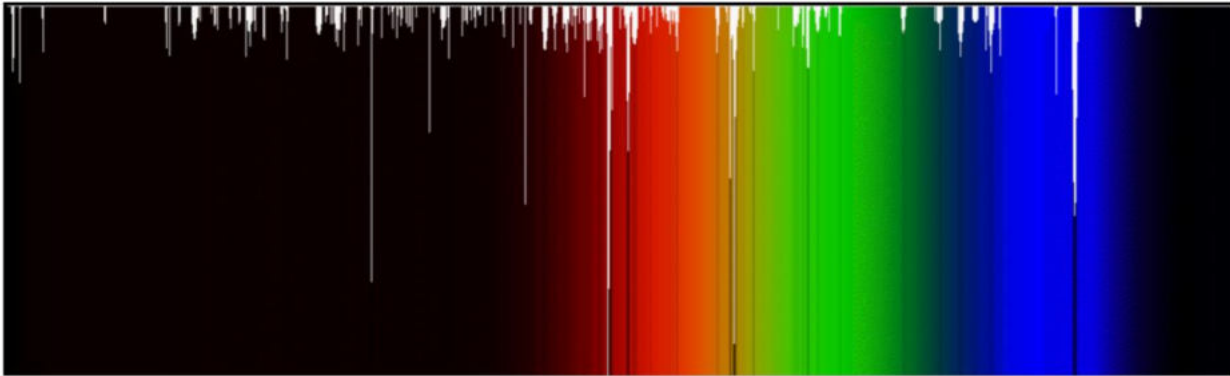
¹ Centre for Astrophysics and Planetary Science, University of Kent, United Kingdom

² Atomic and Molecular Physics Laboratory, Atomki Institute for Nuclear Research, Hungary

³ School of Electronic Engineering and Computer Science, Queen Mary University of London, United Kingdom

January 19, 2022

The Early Years of Astrochemistry (1922-1942)



Feldman, *Can. J. Phys.* 79, 89.

Modern Astrochemistry (1963 – Present)

Molecules are Everywhere!

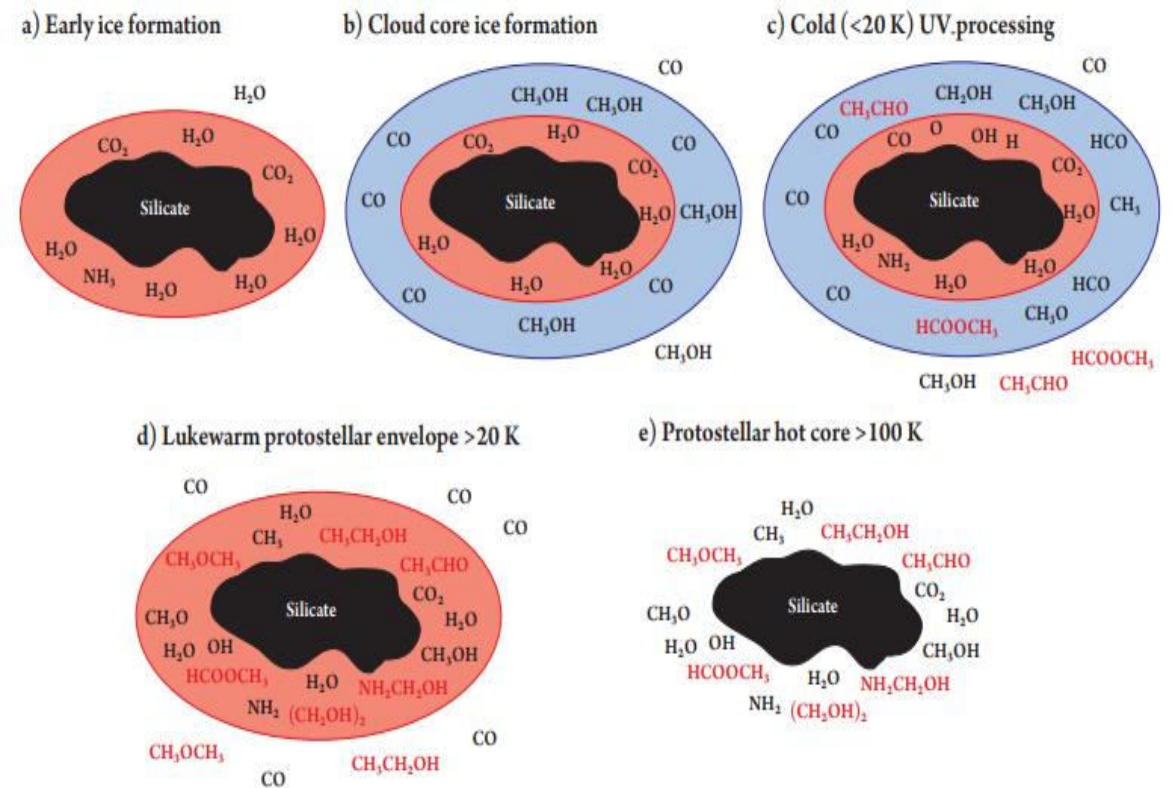
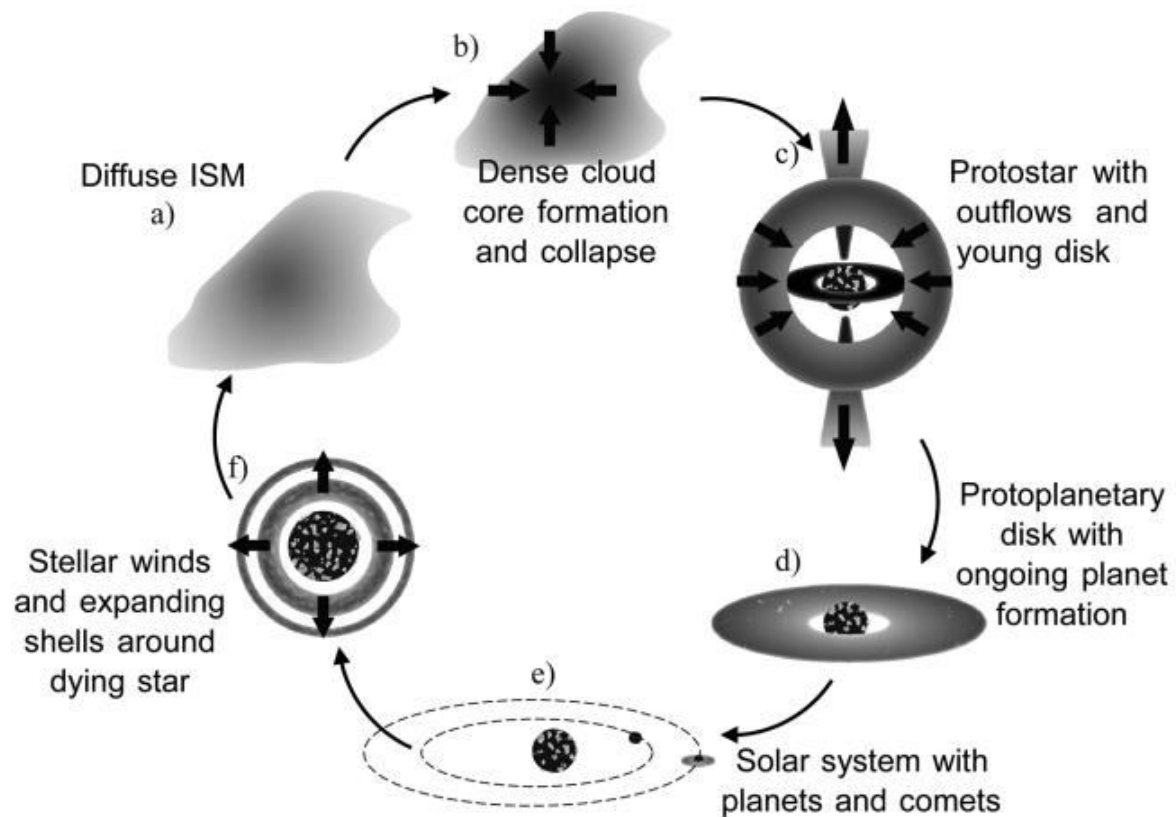
2 Atoms		3 Atoms		4 Atoms		5 Atoms		6 Atoms		7 Atoms		8 Atoms		9 Atoms		10 Atoms		11 Atoms		12 Atoms		13 Atoms		PAHs	Fullerenes
CH	NH	H ₂ O	MgCN	NH ₃	SiC ₃	HC ₃ N	C ₄ H ⁻	CH ₃ OH	CH ₃ CHO	HCOOCH ₃	CH ₃ OCH ₃	CH ₃ COCH ₃	HC ₉ N	C ₆ H ₆	C ₆ H ₅ CN	1-C ₁₀ H ₇ CN	C ₆₀								
CN	SiN	HCO ⁺	H ₃ ⁺	H ₂ CO	CH ₃	HCOOH	CNCHO	CH ₃ CN	CH ₃ CCH	CH ₃ C ₃ N	CH ₃ CH ₂ OH	HOCH ₂ CH ₂ OH	CH ₃ C ₆ H	n-C ₃ H ₇ CN	HC ₁₁ N	2-C ₁₀ H ₇ CN	C ₆₀ ⁺								
CH ⁺	SO ⁺	HCN	SiCN	HNCO	C ₃ N ⁻	CH ₂ NH	HNCNH	NH ₂ CHO	CH ₃ NH ₂	C ₇ H	CH ₃ CH ₂ CN	CH ₃ CH ₂ CHO	C ₂ H ₅ OCHO	i-C ₃ H ₇ CN		C ₉ H ₈	C ₇₀								
OH	CO ⁺	OCS	AlNC	H ₂ CS	PH ₃	NH ₂ CN	CH ₃ O	CH ₃ SH	CH ₂ CHCN	CH ₃ COOH	HC ₇ N	CH ₃ C ₅ N	CH ₃ COOCH ₃	1-C ₅ H ₅ CN											
CO	HF	HNC	SiNC	C ₂ H ₂	HCNO	H ₂ CCO	NH ₃ D ⁺	C ₂ H ₄	HC ₅ N	H ₂ C ₆	CH ₃ C ₄ H	CH ₃ CHCH ₂ O	CH ₃ COCH ₂ OH	2-C ₅ H ₅ CN											
H ₂	N ₂	H ₂ S	HCP	C ₃ N	HOCN	C ₄ H	H ₂ NCO ⁺	C ₅ H	C ₆ H	CH ₂ OHCHO	C ₈ H	CH ₃ OCH ₂ OH	C ₅ H ₆												
SiO	CF ⁺	N ₂ H ⁺	CCP	HNCS	HSCN	SiH ₄	NCCNH ⁺	CH ₃ NC	c-C ₂ H ₄ O	HC ₆ H	CH ₃ CONH ₂														
CS	PO	C ₂ H	AlOH	HOCO ⁺	HOOH	c-C ₃ H ₂	CH ₃ Cl	HC ₂ CHO	CH ₂ CHOH	CH ₂ CHCHO	C ₈ H ⁻														
SO	O ₂	SO ₂	H ₂ O ⁺	C ₃ O	l-C ₃ H ⁺	CH ₂ CN	MgC ₃ N	H ₂ C ₄	C ₆ H ⁻	CH ₂ CCHCN	CH ₂ CHCH ₃														
SiS	AlO	HCO	H ₂ Cl ⁺	l-C ₃ H	HMgNC	C ₅	HC ₃ O ⁺	C ₅ S	CH ₃ NCO	NH ₂ CH ₂ CN	CH ₃ CH ₂ SH														
NS	CN ⁻	HNO	KCN	HCNH ⁺	HCCO	SiC ₄	NH ₂ OH	HC ₃ NH ⁺	HC ₅ O	CH ₃ CHNH	HC ₇ O														
C ₂	OH ⁺	HCS ⁺	FeCN	H ₃ O ⁺	CNCN	H ₂ CCC	HC ₃ S ⁺	C ₅ N	HOCH ₂ CN	CH ₃ SiH ₃	CH ₃ NHCHO														
NO	SH ⁺	HOC ⁺	HO ₂	C ₃ S	HONO	CH ₄	H ₂ CCS	HC ₄ H	HC ₄ NC	NH ₂ CONH ₂	H ₂ CCCHCCH														
HCl	HCl ⁺	SiC ₂	TiO ₂	c-C ₃ H	MgCCH	HCCNC	C ₄ S	HC ₄ N	HC ₃ HNH	HCCCH ₂ CN	HCCCHCHCN														
NaCl	SH	C ₂ S	CCN	HC ₂ N	HCCS	HNCCC	CHOSH	c-H ₂ C ₃ O	c-C ₃ HCCH	CH ₂ CHCCH	H ₂ CCHC ₃ N														
AlCl	TiO	C ₃	SiCSi	H ₂ CN		H ₂ COH ⁺		CH ₂ CNH																	
KCl	ArH ⁺	CO ₂	S ₂ H					C ₅ N ⁻																	
AlF	NS ⁺	CH ₂	HCS					HNCHCN																	
PN	HeH ⁺	C ₂ O	HSC					SiH ₃ CN																	
SiC	VO	MgNC	NCO					MgC ₄ H																	
CP		NH ₂	CaNC					CH ₃ CO ⁺																	
		NaCN	NCS					H ₂ CCCS																	
		N ₂ O						CH ₂ CCH																	

High Complexity

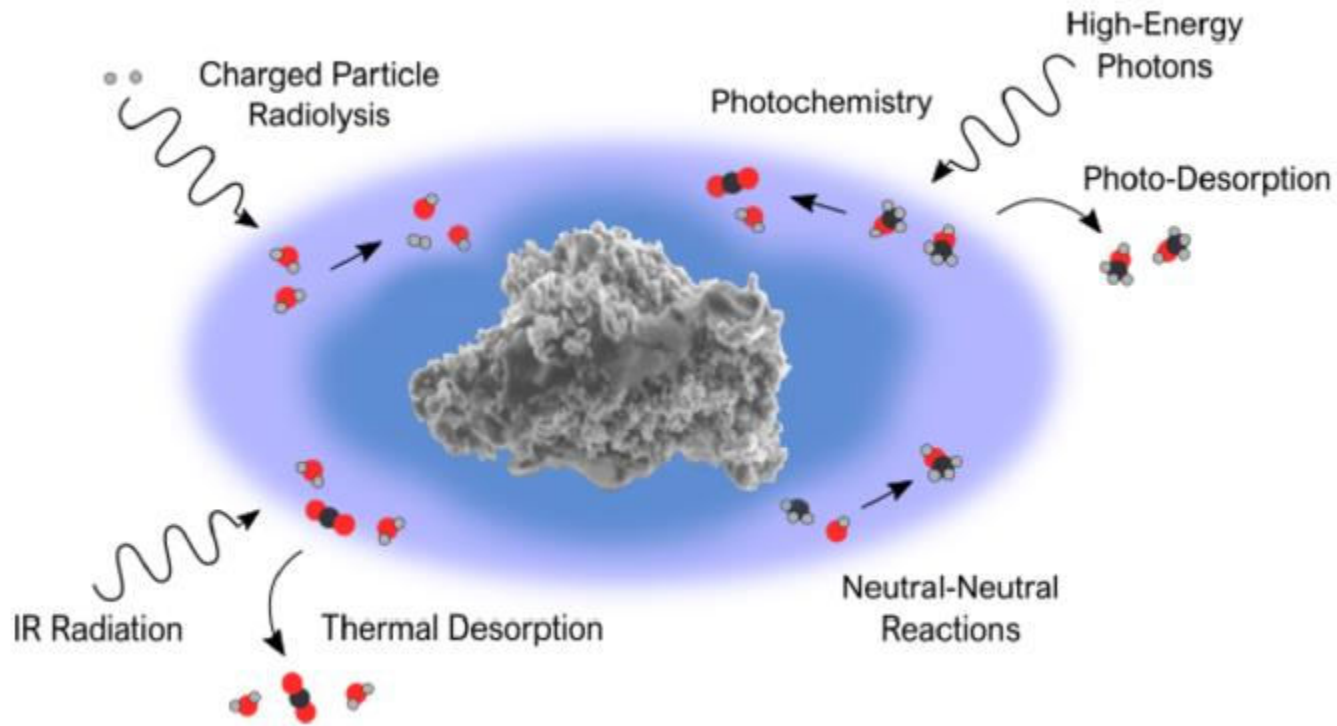
Biologically Relevant

Mineralogically Relevant

Cosmic Chemistry Cycle



Ices – the Factories of Interstellar Molecular Complexity!

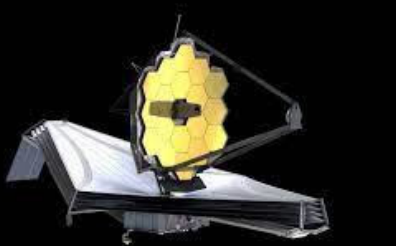
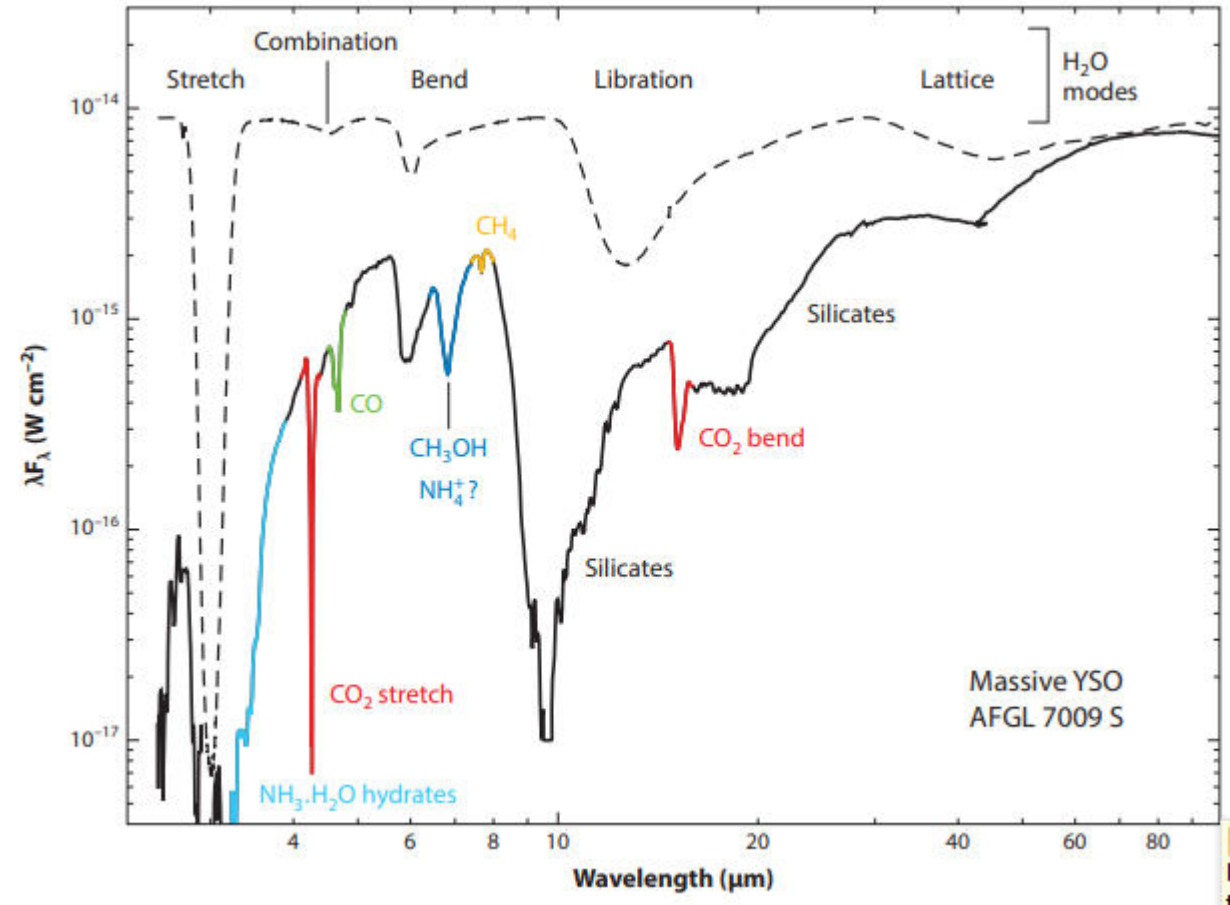
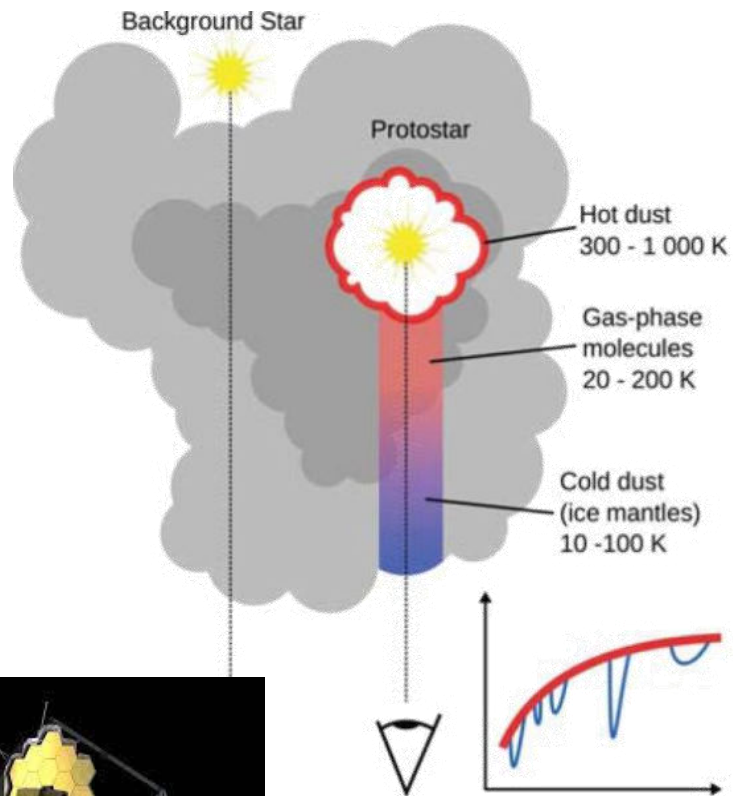


Herczku et al., *Rev. Sci. Instrum.* **92**, 084501.

Species	$X_{\text{H}_2\text{O}}^a$ [%]			
	MYSOs	LYSOs	BG Stars ^c	Comets
Securely identified species: ^d				
H_2O^e	100	100	100	100
CO^e	7_4^{15} (7) 3–26	21_{12}^{35} (18) (<3)–85	25_{20}^{43} 9–67	nd 0.4–30
CO_2^e	19_{12}^{25} 11–27	28_{23}^{37} 12–50	26_{18}^{39} 14–43	15_{10}^{24} 4–30
CH_3OH	9_5^{23} (5) (<3)–31	6_5^{12} (5) (<1)–25	8_6^{10} (6) (<1)–12	nd 0.2–7
NH_3	nd $\sim 7^f$	6_4^8 (4) 3–10	nd <7	nd 0.2–1.4
CH_4	nd 1–3	4.5_3^6 (3) 1–11	nd <3	nd 0.4–1.6
Likely identified species: ^g				
H_2CO	~ 2 –7	~ 6	nd	0.11–1.0
OCN^-	$0.6_{0.3}^{0.7}$ 0.1–1.9	$0.6_{0.4}^{0.8}$ (0.4) (<0.1)–1.1	nd <0.5	nd nd
OCS	0.03–0.16	≤ 1.6	<0.22	0.1–0.4

Boogert et al., *Annu. Rev. Astron. Astrophys.* **53**, 541.

Constraining the Chemical Compositions of Icy Mantles



Constraining the Chemical Compositions of Icy Mantles

- Several telescopes available to collect mid-IR and THz spectra
- Laboratory spectral data needed to interpret observations



Advantages of THz Spectroscopy Compared to Mid-IR

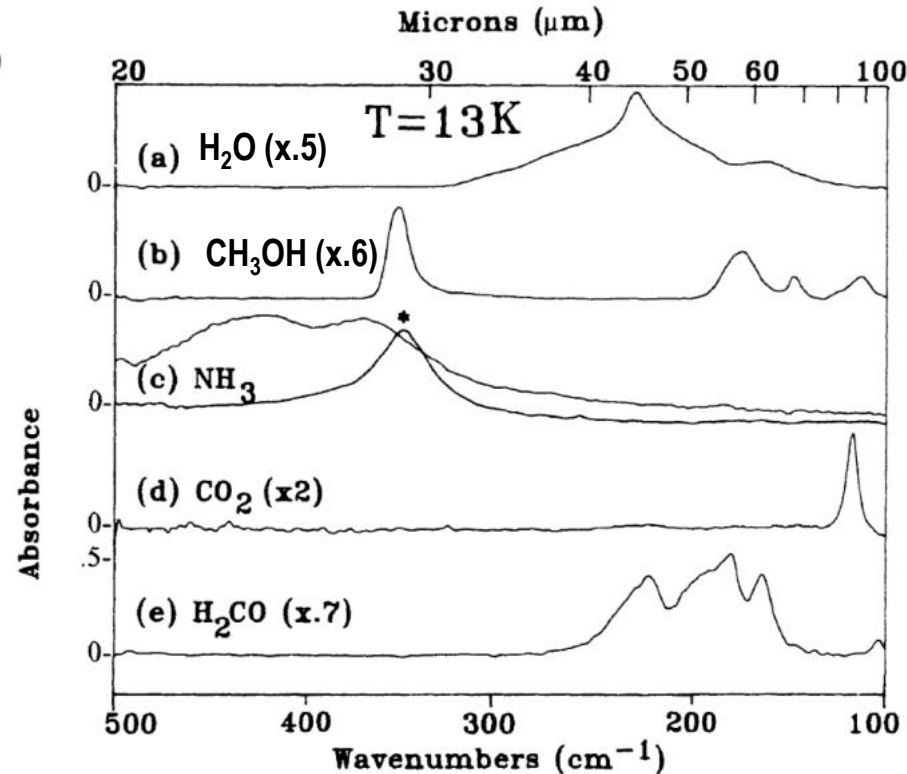
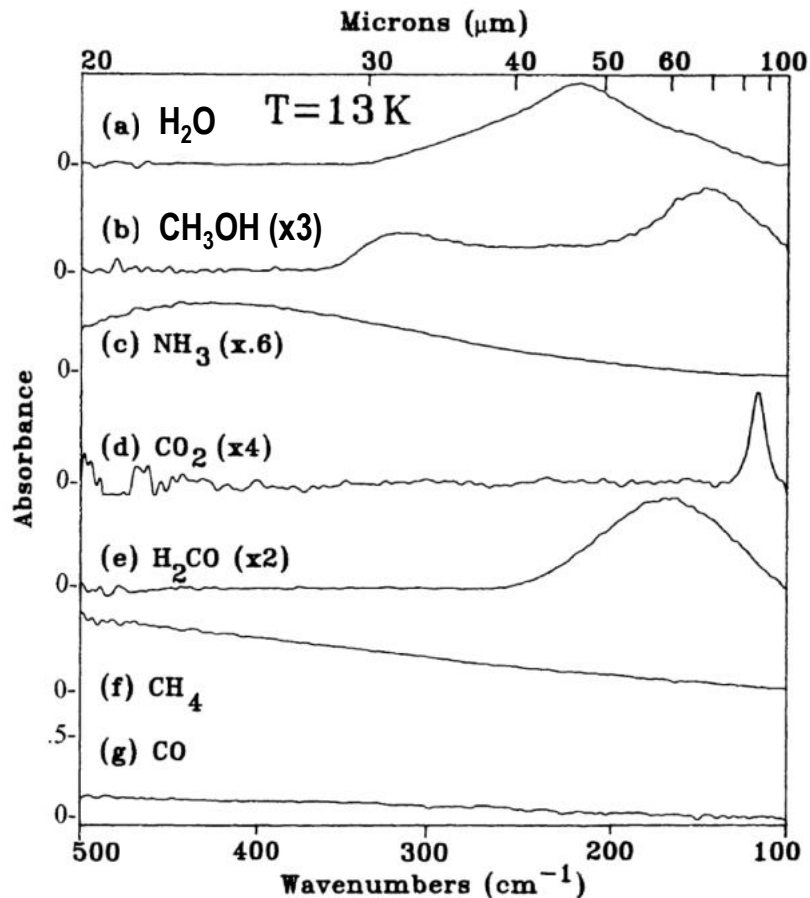
- **Mid-IR**
 - Only intra-molecular vibrational modes can be observed
 - Different molecules sharing functional moieties may be hard to identify
- **THz**
 - Lower energy vibrations are probed, including inter-molecular ones
 - Technique is sensitive to long-range interactions between molecules and ice structure

Advantages of THz Spectroscopy Compared to Mid-IR

- **Mid-IR**
 - Transitions can only be observed in absorption
 - A mid-IR source (e.g., a YSO) behind the line of sight is needed, greatly reducing the number of astronomical settings that may be studied
- **THz**
 - Low energetic transitions mean that at <150 K absorption can occur against the background continuum
 - Molecules may therefore be studied via their emission signals

Early Forays into Laboratory Far-IR Spectroscopy

10

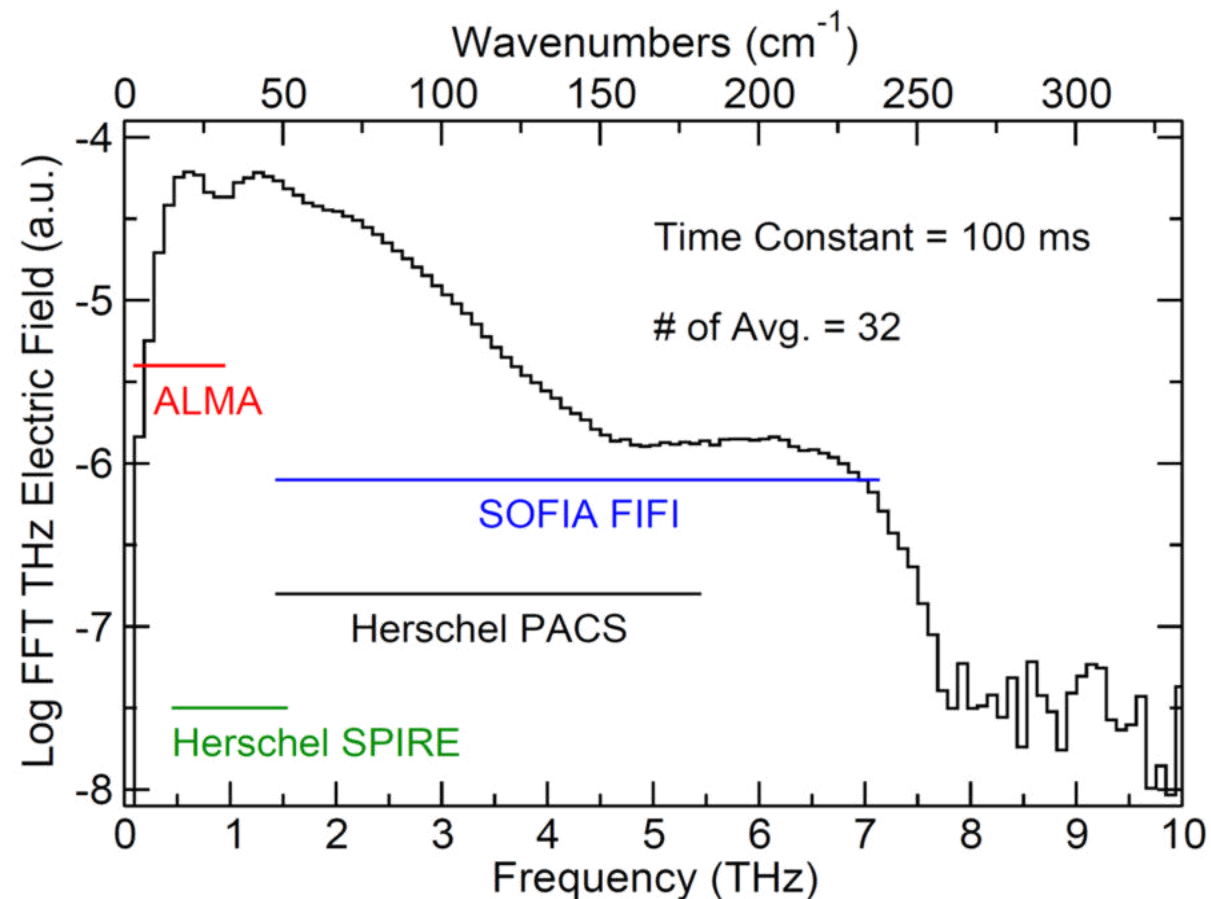
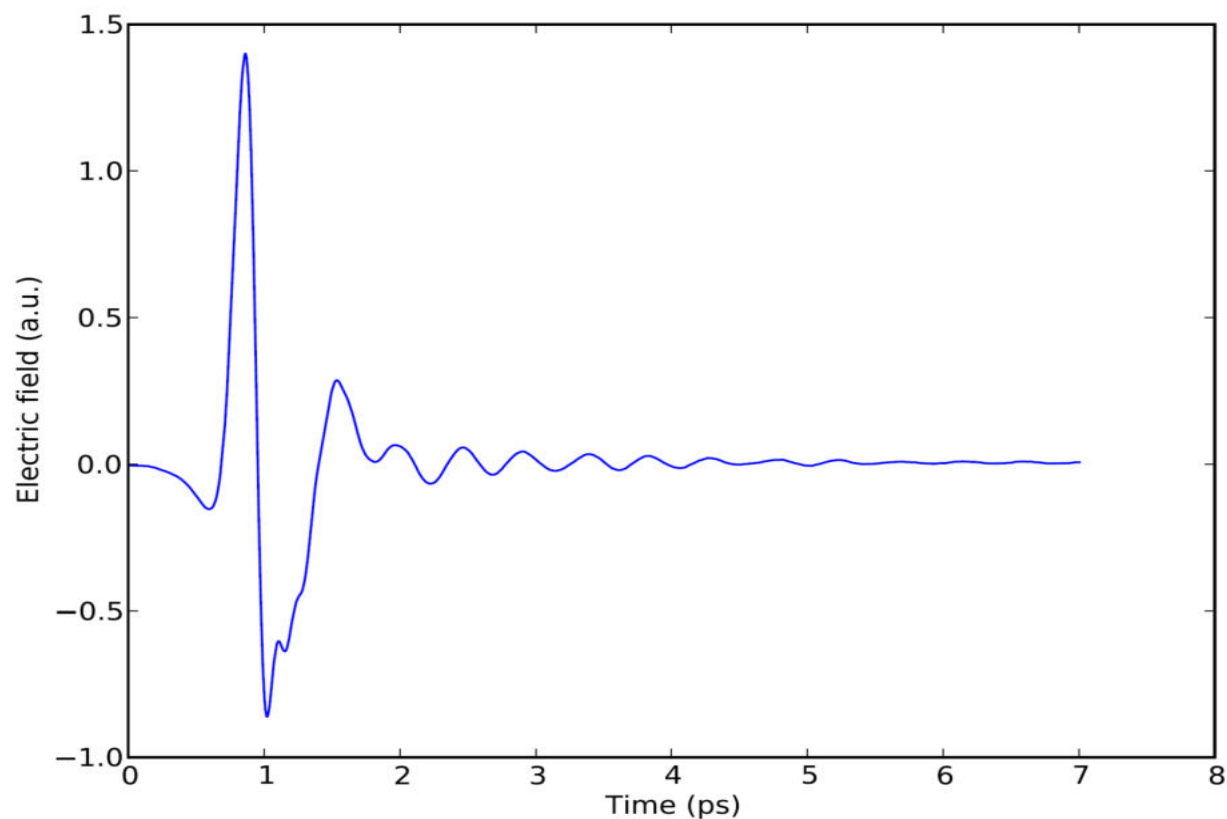


- Noticeable differences in the spectra of amorphous (left) and crystalline (right) ices.
- Indication of the usefulness of THz spectroscopy.

Time-Domain (TD) THz Spectroscopy in the Laboratory

An Example from Caltech

Collect a signal in the time domain.....and Fourier transform numerically

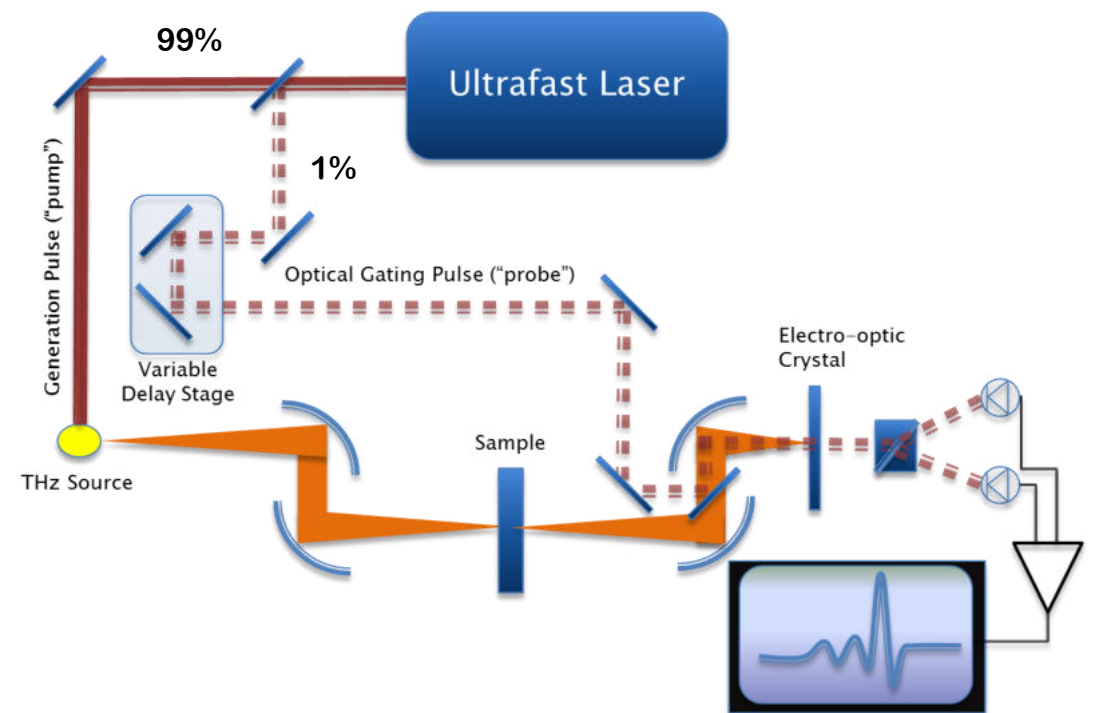


Time-Domain (TD) THz Spectroscopy in the Laboratory

An Example from Caltech

Laboratory TD-THz spectroscopy can be broken down into two very basic concepts:

- THz Pulse Generation
 - *via two-colour plasma*
- THz Detection
 - *via electro-optical sampling*



Time-Domain (TD) THz Spectroscopy in the Laboratory

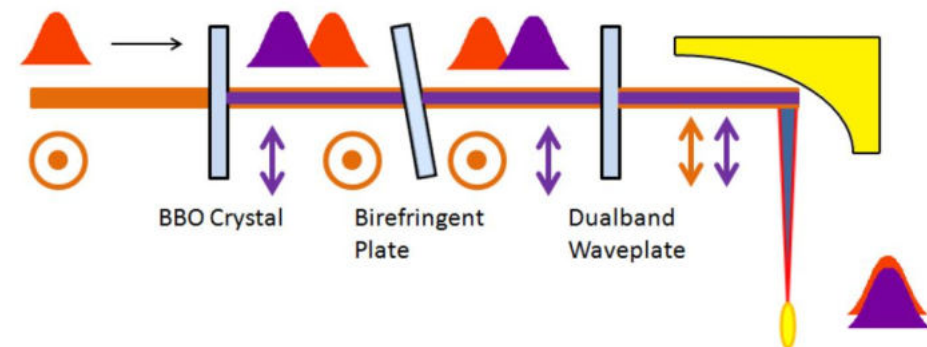
An Example from Caltech

THz Pulse Generation

- BBO – frequency double amplified 800 nm output to 400 nm
- Calcite – compensate for phase delay between 400 and 800 nm pulses
- Dual-Band Wave-Plate – align polarisations
- Focus 400 and residual 800 nm light to form plasma in air

Mechanism:

- 800 nm light generates plasma
- 400 nm light accelerates electrons in plasma
- Accelerated electrons emit THz pulses



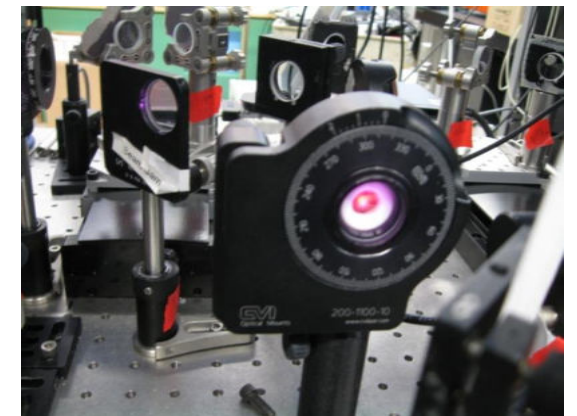
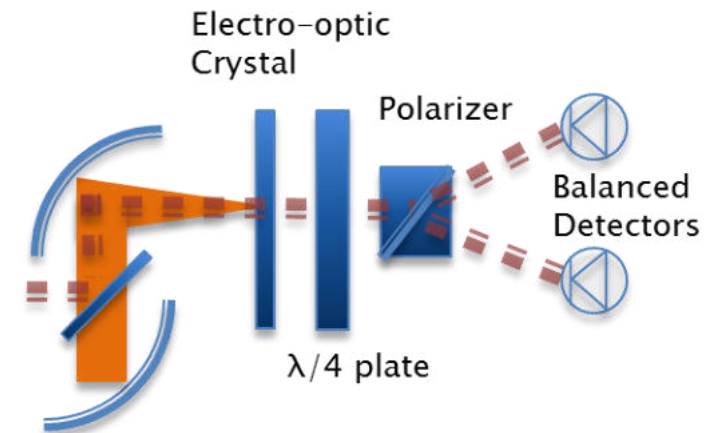
Time-Domain (TD) THz Spectroscopy in the Laboratory

An Example from Caltech

14

THz Detection

- THz beams are focused down onto a crystal (ZnTe or GaP)
- THz pulse causes a rotation in the polarisation of the probe beam in the crystal (Pockels Effect)
- Magnitude of polarisation change is linear to applied THz electric field (so measure THz electric field not intensity)
- A pair of balanced detectors sees a difference in signal

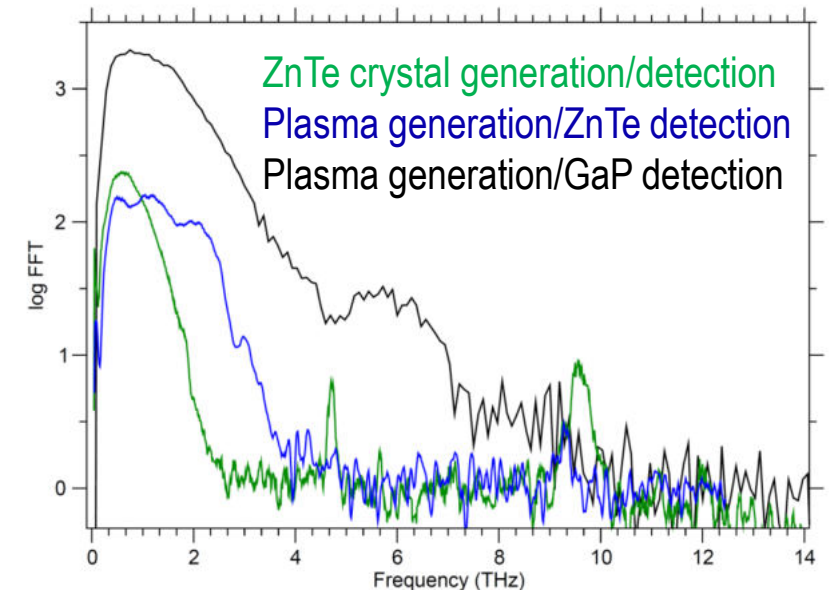
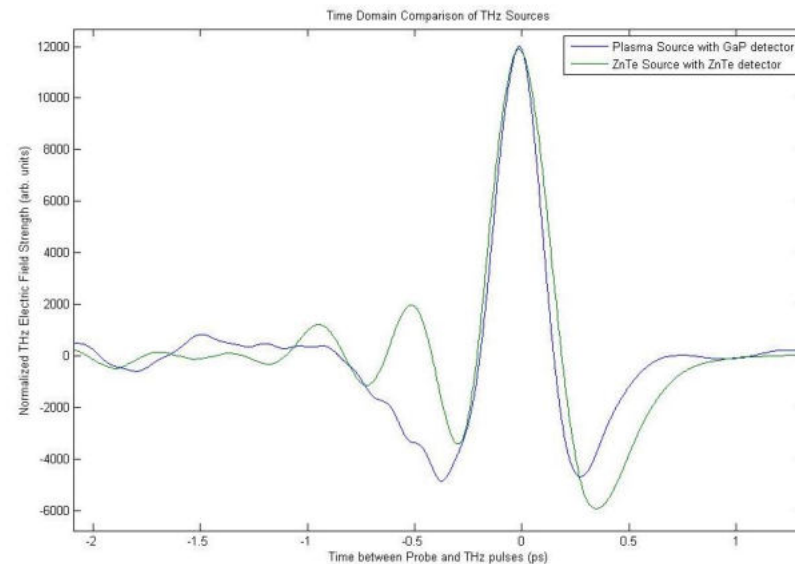
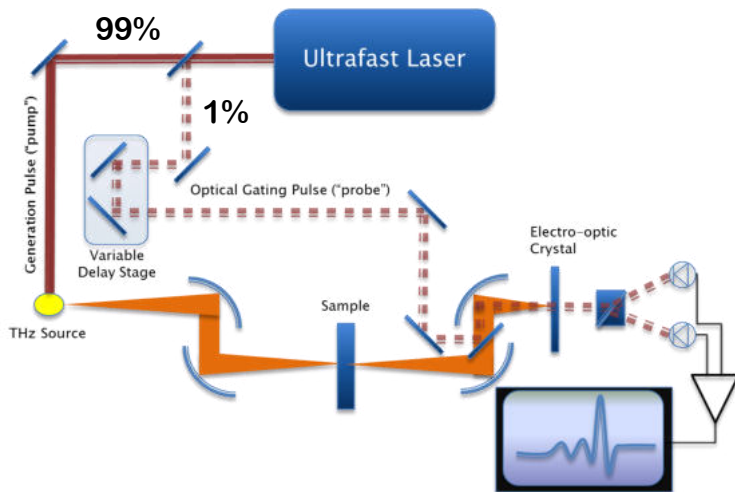


Time-Domain (TD) THz Spectroscopy in the Laboratory

An Example from Caltech

Opto-Mechanical Delay

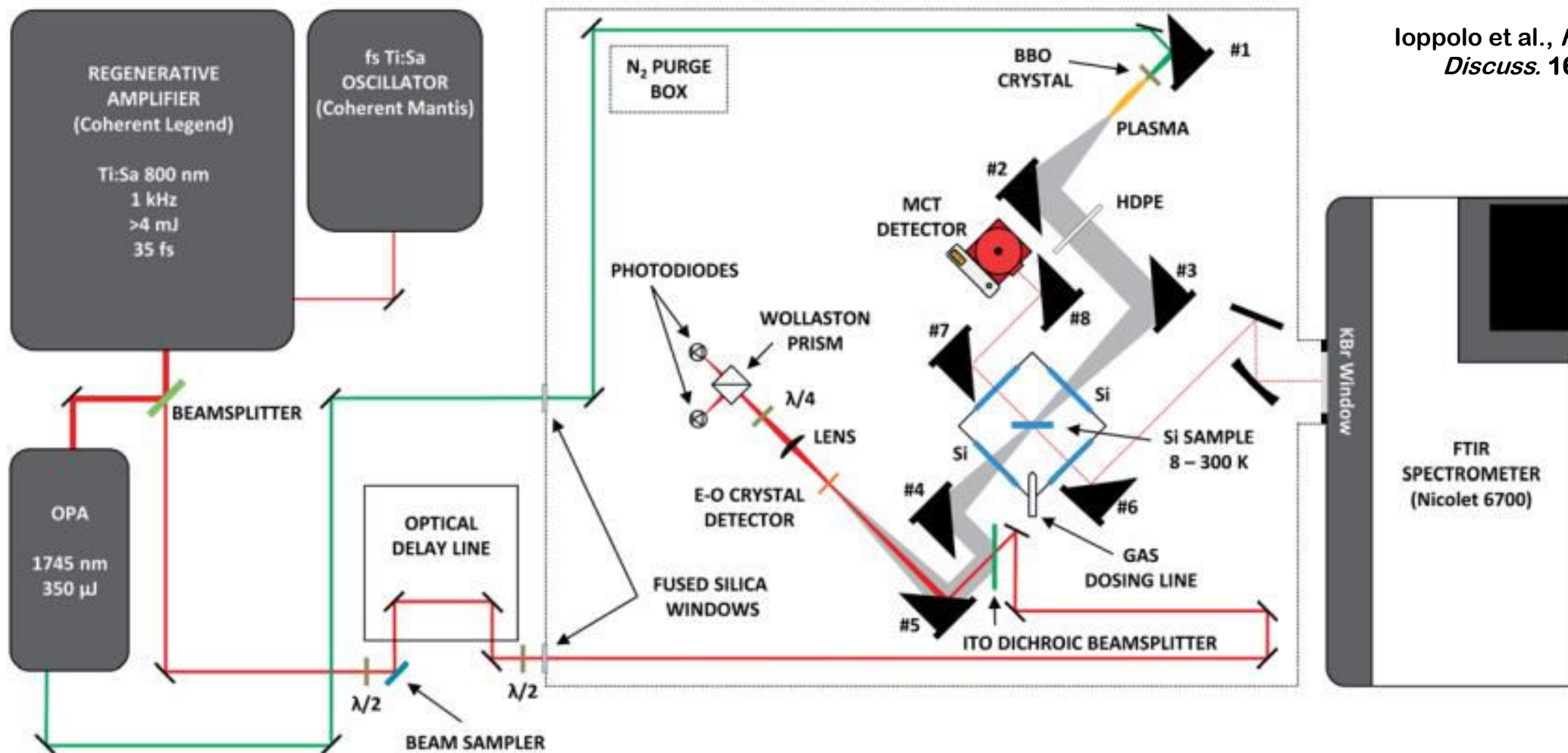
- Allows the entire THz waveform to be stepped through
- The electric field is measured as a function of delay time
- FFT of the temporal waves gives spectral distribution in the frequency domain.



Time-Domain (TD) THz Spectroscopy in the Laboratory

An Example from Caltech

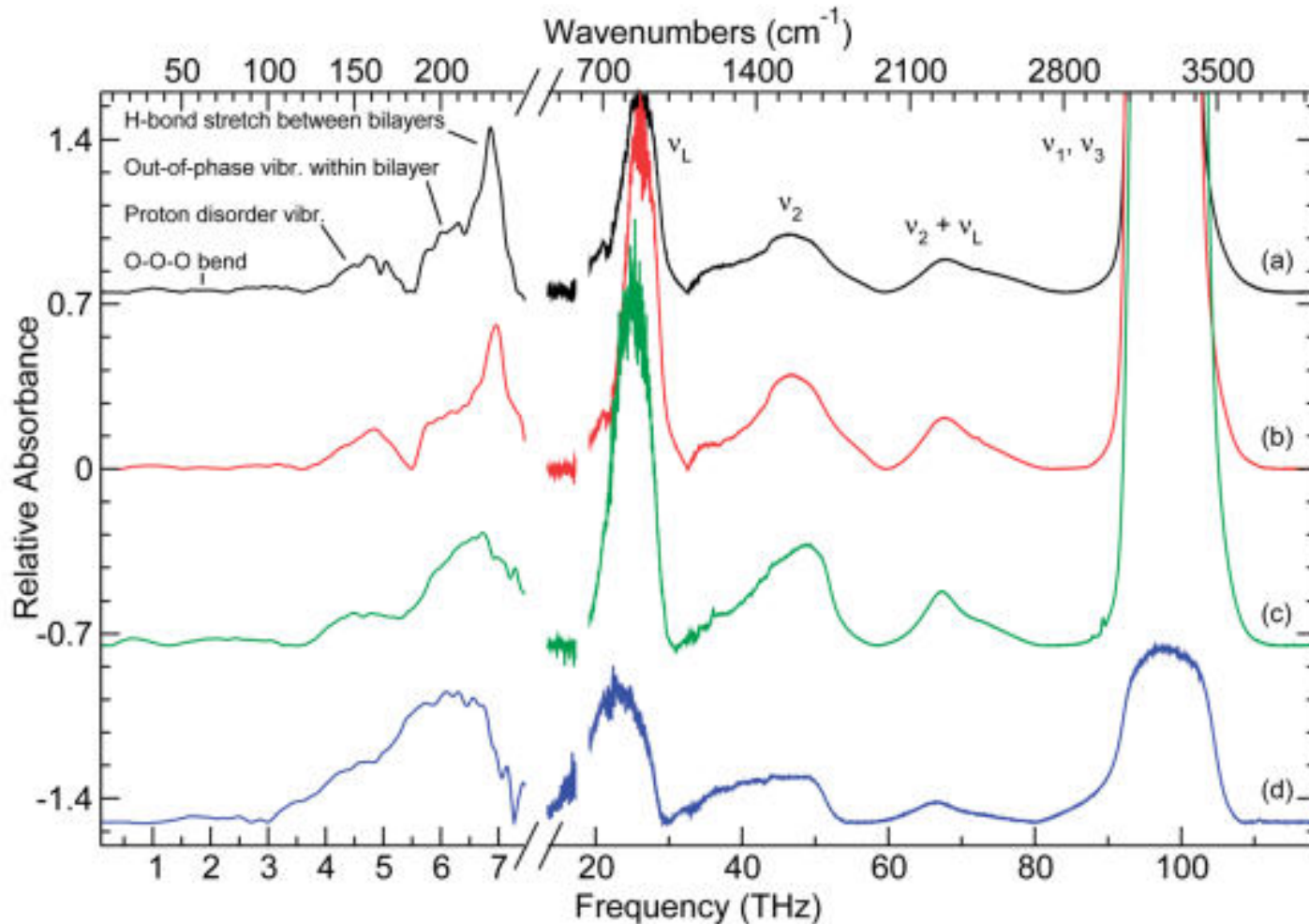
16



Ioppolo et al., *Faraday Discuss.* 168, 461.

THz Spectra of Astrophysical Ice Analogues

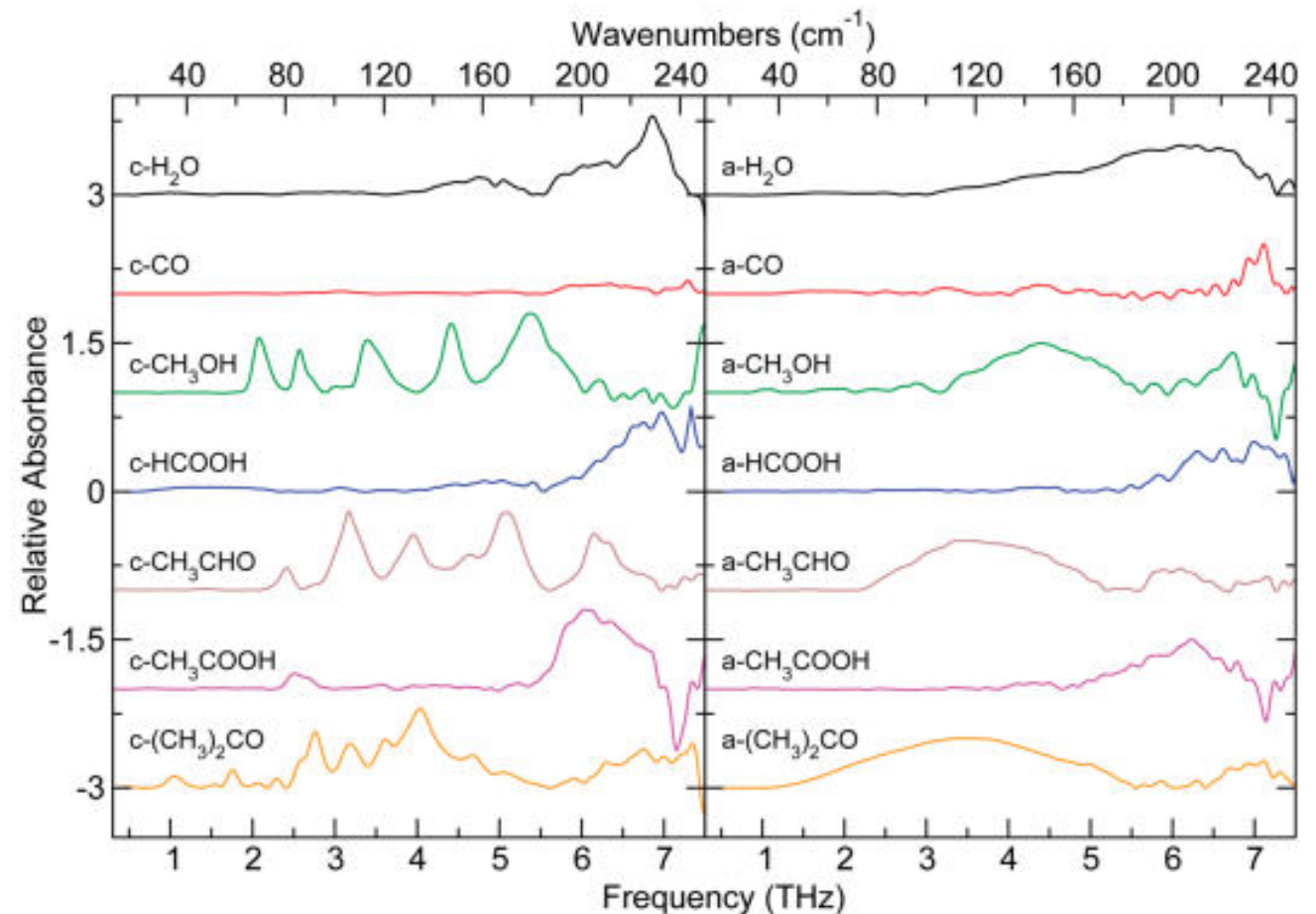
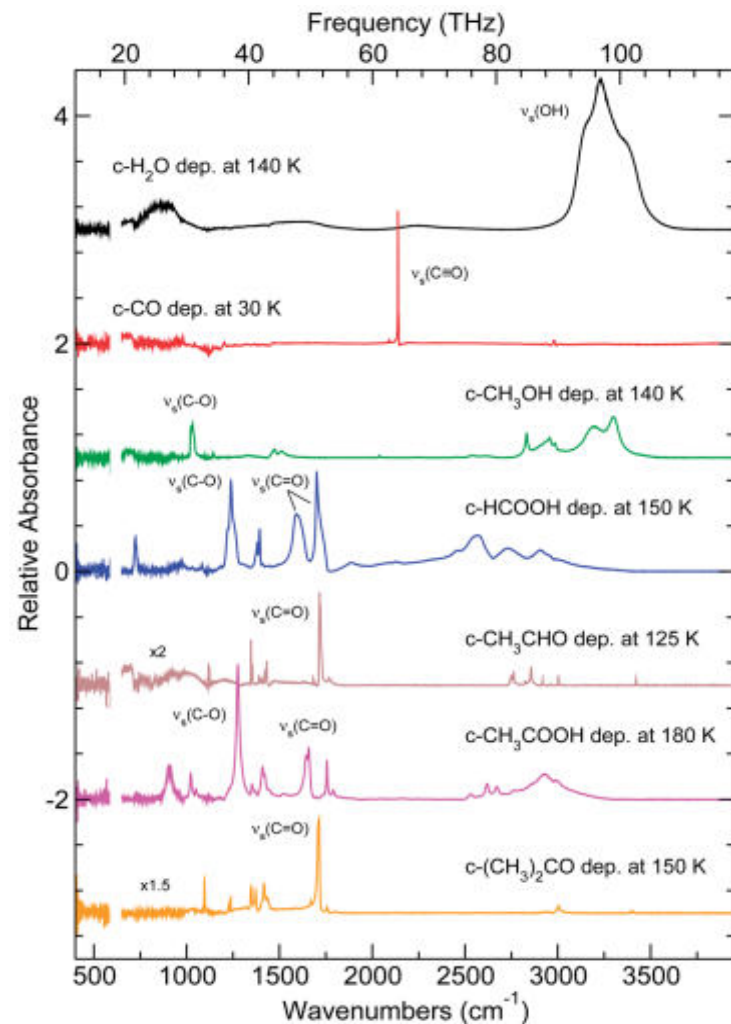
Comparisons with Mid-IR Spectra for H₂O



- Four absorptions in the H₂O mid-IR spectrum
- Peak appearances vary due to temperature and phase.
- THz spectra contains many weak features.
- Four strong absorptions in crystalline H₂O.

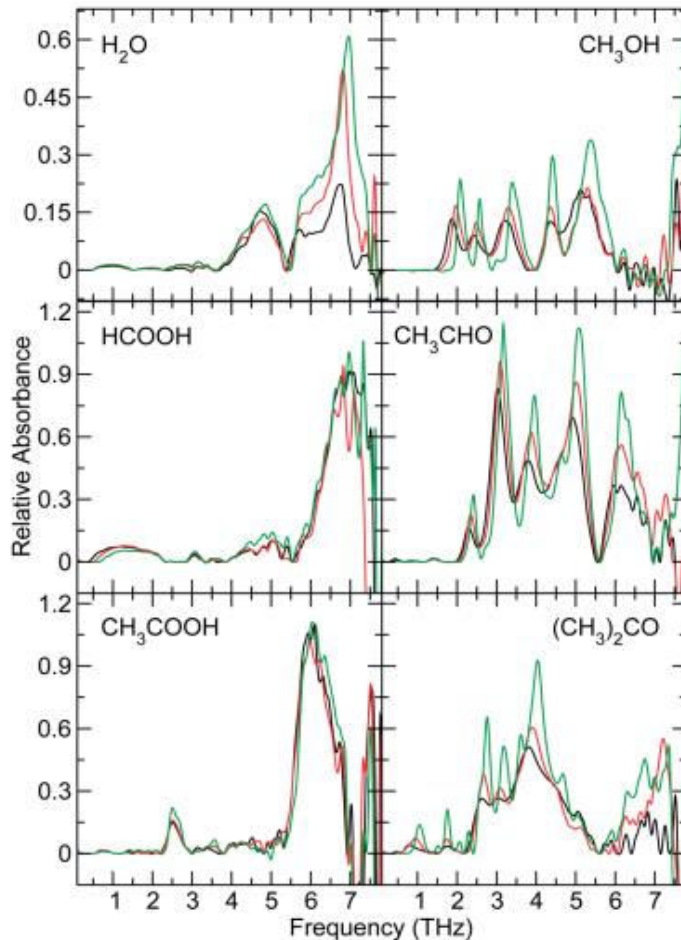
THz Spectra of Astrophysical Ice Analogues

Comparisons with Mid-IR Spectra for Other Molecules



THz Spectra of Astrophysical Ice Analogues

The Temperature Effect

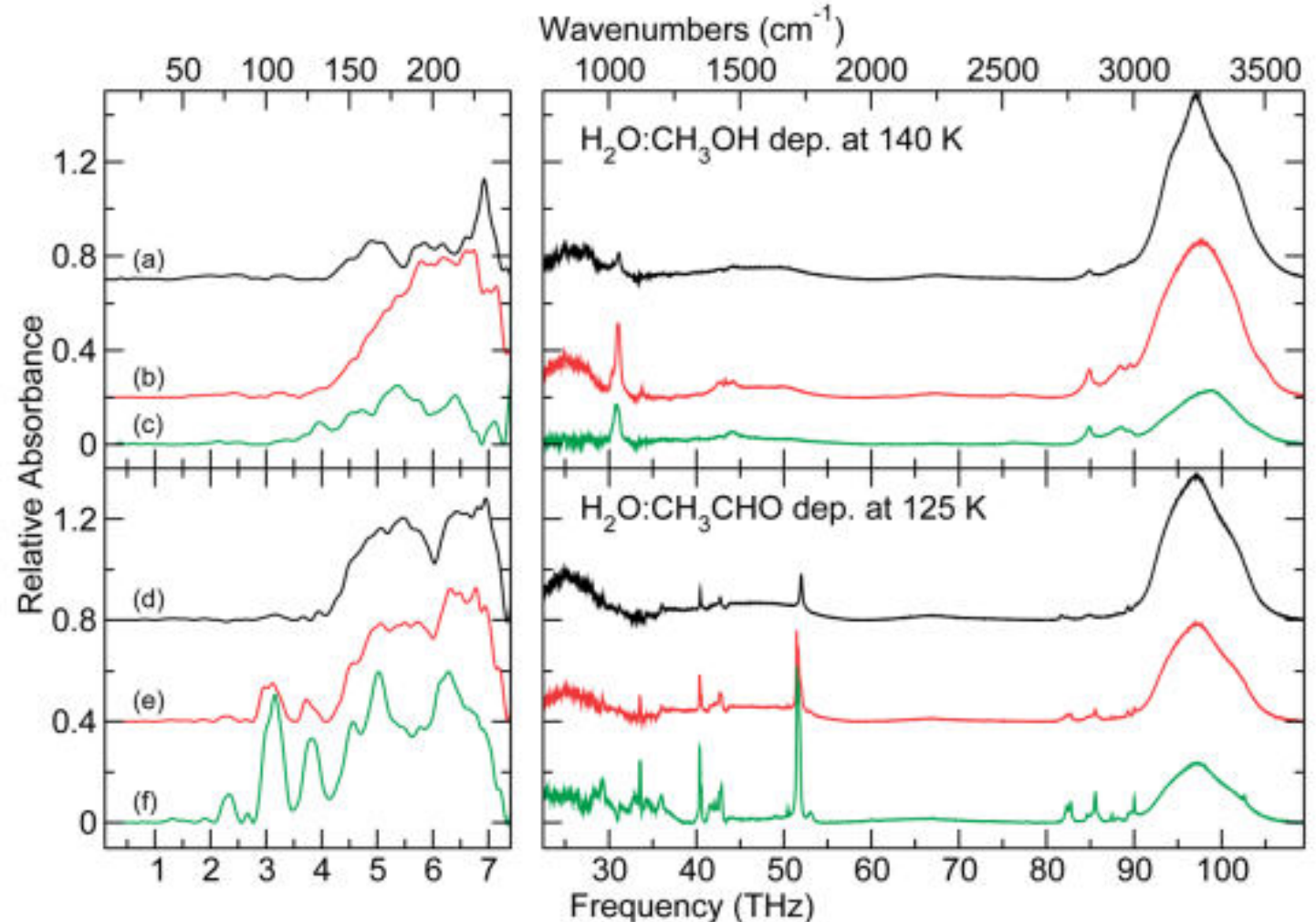


- THz peaks in crystalline ices are stronger, sharper, and blue-shifted at lower temperatures.
- Peak shifting and broadening results from the anharmonicity of vibrational potential.
- Hot bands are red-shifted due to decrease in the spacing between vibrational levels.
- At low temperature, only lower vibrational states are populated and so bands are blue-shifted and sharper.
- Different vibrational minima corresponding to different crystal structures will exhibit different bands.
- Intramolecular torsional modes may be apparent.

THz Spectra of Astrophysical Ice Analogues

The Composition Effect

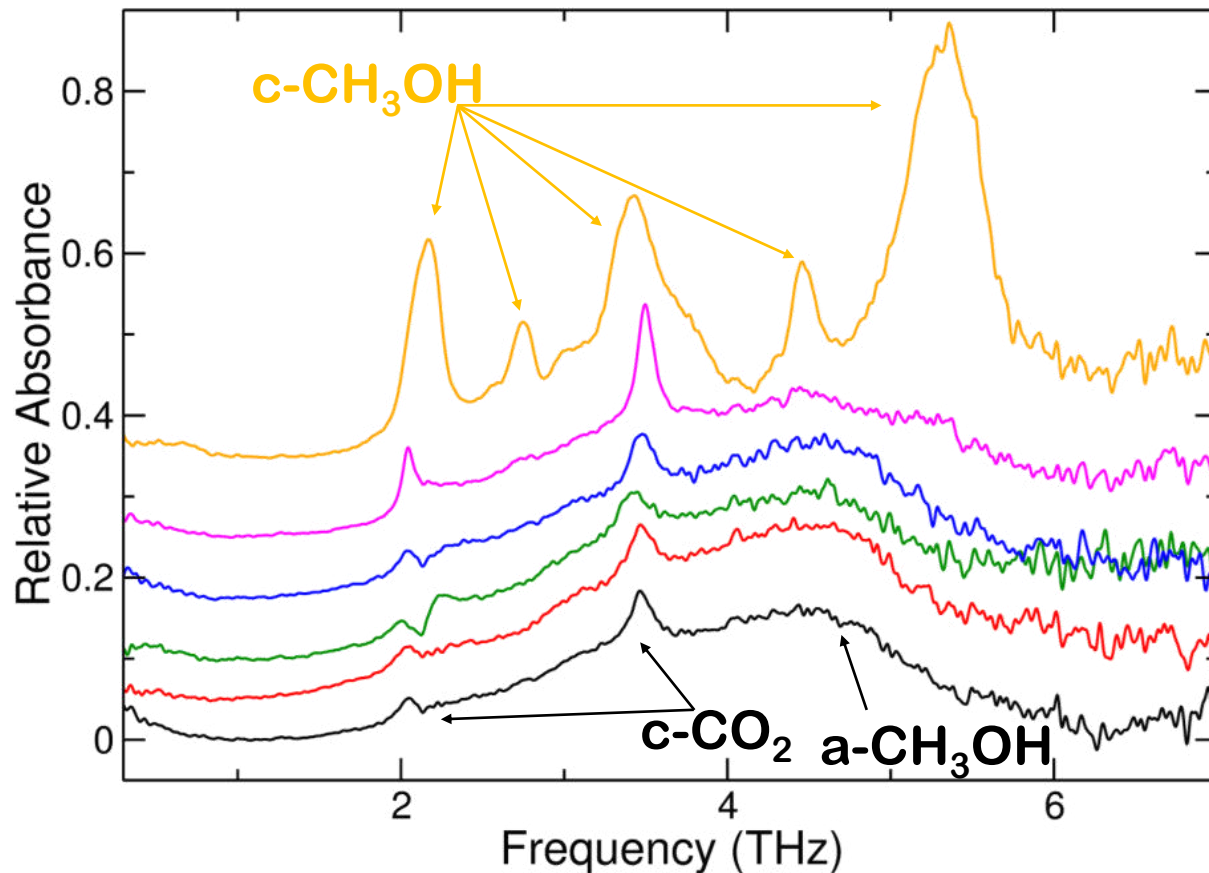
- Contaminant species may alter the intermolecular H-bonding networks in ices.
- If the peaks of a contaminant affect those of the bulk ice independent of contaminant concentration, and these peaks also grow proportionally to their concentration, then segregation may have occurred (e.g., CO₂, CH₃CHO).



THz Spectra of Astrophysical Ice Analogues

The Composition Effect

Thermal processing of mixed $\text{CH}_3\text{OH}:\text{CO}_2$ 1:1 ice

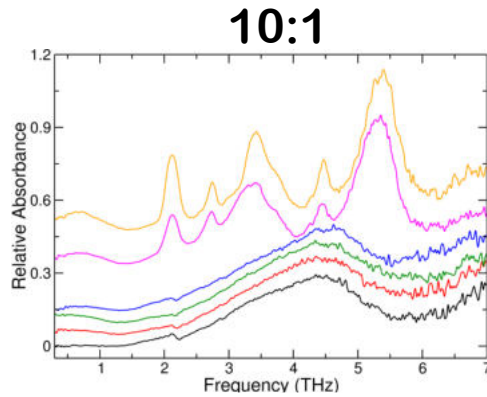


McGuire et al., *Phys. Chem. Chem. Phys.* **18**, 20199.

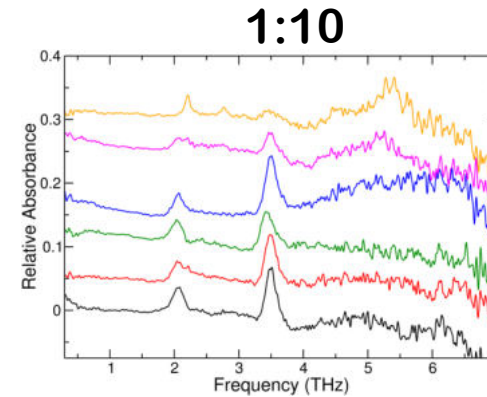
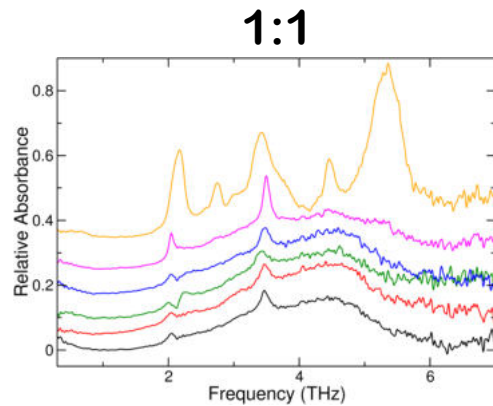
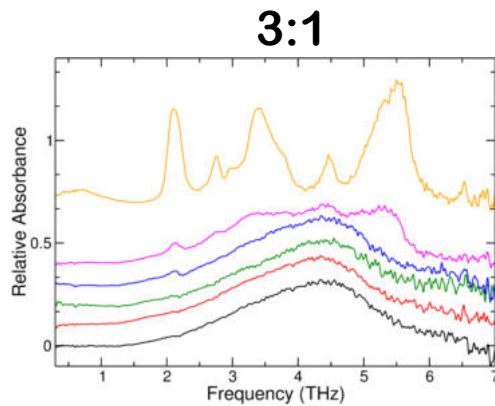
- Ann. to 140 K, ↓ 10 K
- Ann. to 120 K, ↓ 10 K
- Ann. to 90 K, ↓ 10 K
- Spectrum @ 60 K
- Spectrum @ 30 K
- Deposited @ 10 K

THz Spectra of Astrophysical Ice Analogues

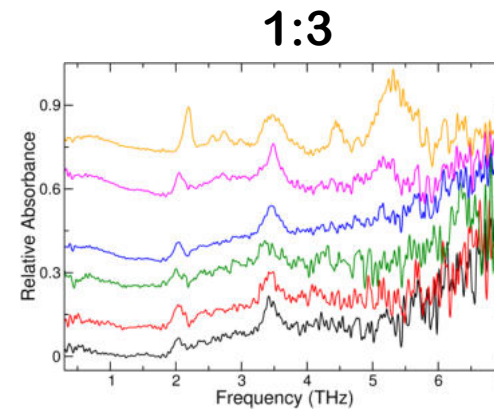
The Composition Effect



CH₃OH >>



<<CO₂



McGuire et al., *Phys. Chem. Chem. Phys.* **18**, 20199.

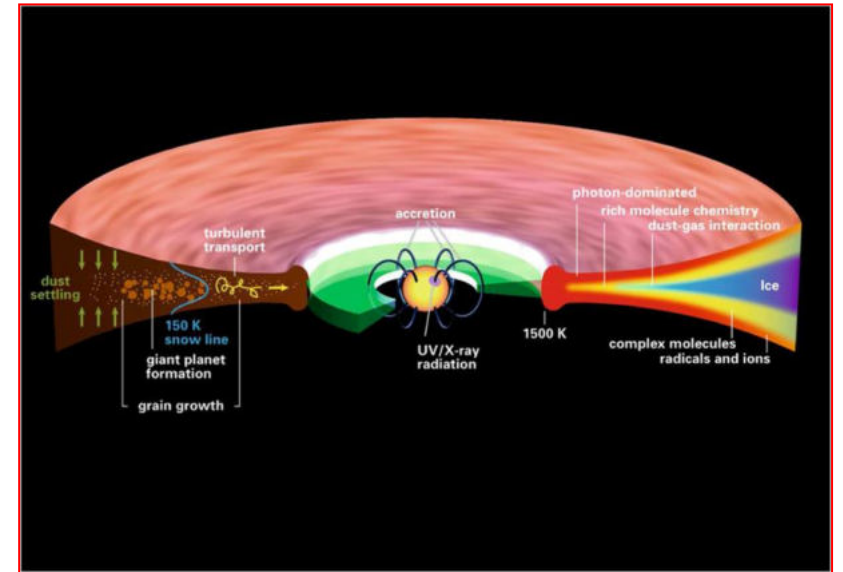
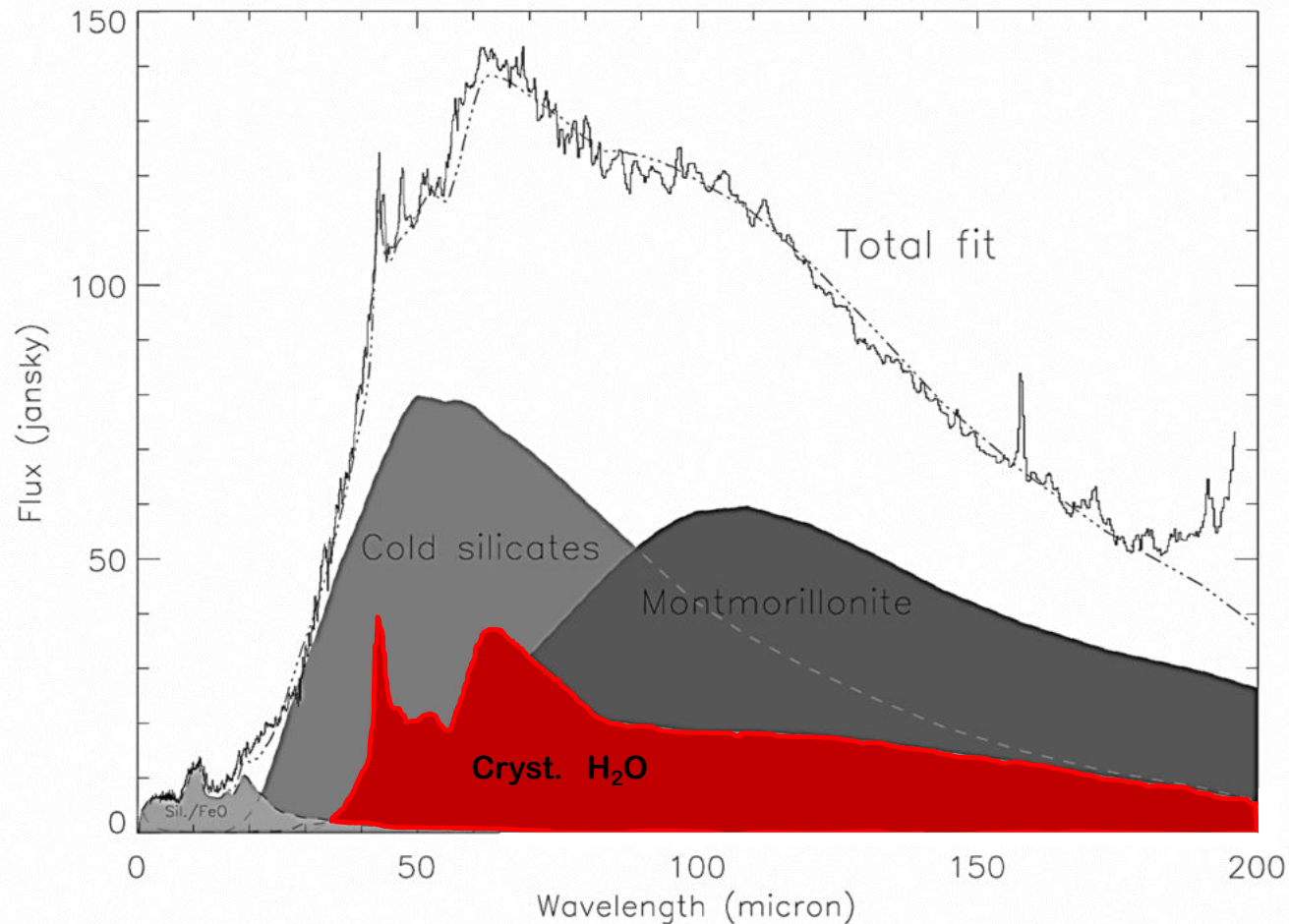
Key Points of THz Laboratory Astrochemistry

23

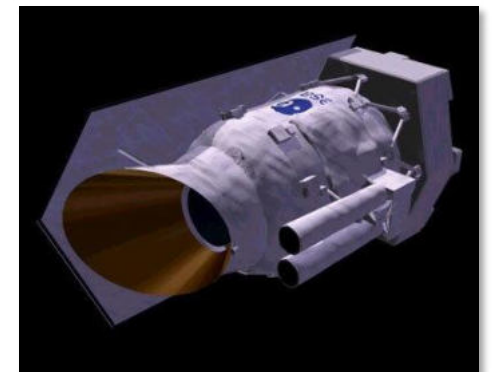
- **THz spectroscopy offers an additional and complementary analytical tool to mid-IR spectroscopy.**
- **THz spectroscopy is not limited to a particular line of sight, offering a wider choice of observable targets.**
- **THz spectroscopy is sensitive to the temperature, structure, and chemical composition of an astrophysical ice.**
- **Further laboratory spectra must be acquired to aid in deciphering astrophysical observations.**

Detection of Ices in Protoplanetary Disks

ISO Data - Herbig Ae Stars HD142527

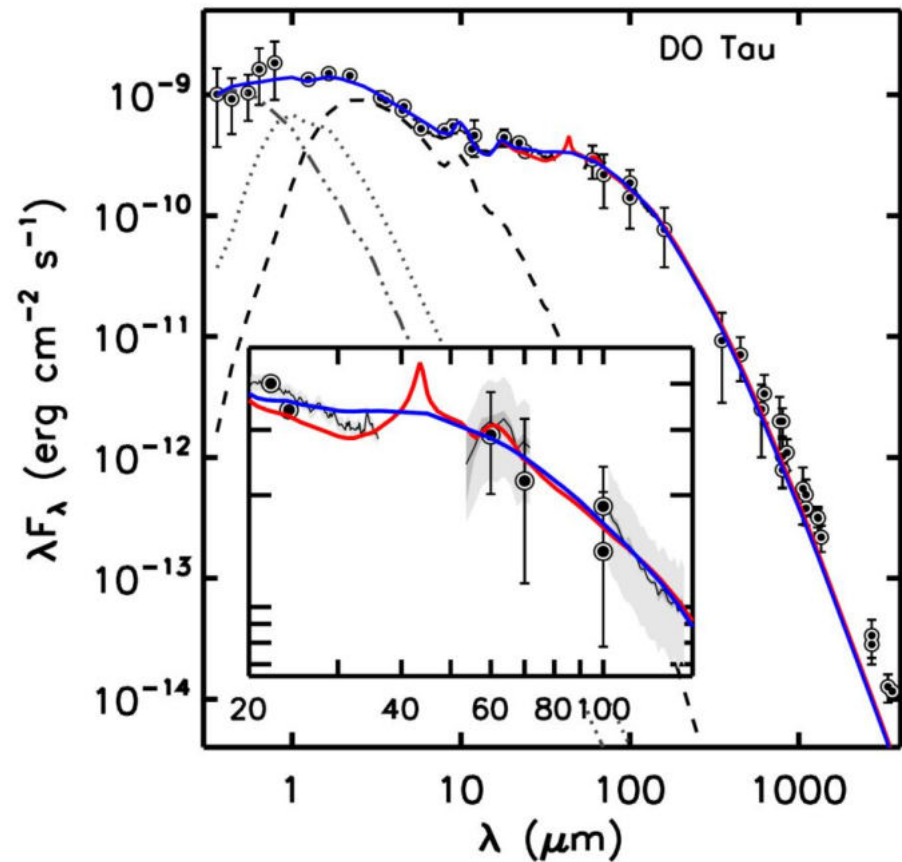


Malfait et al. (1999)



Detection of Trans-Neptunian Ices

PACS Data - T Tauri Stars



with water ice

with silicates and graphite

McClure, PhD Thesis (2014)

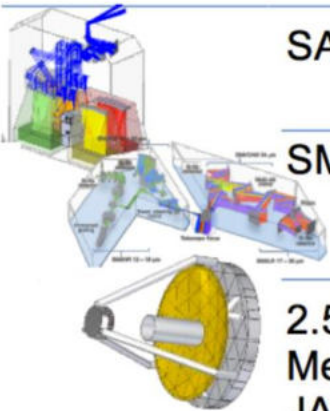


A Missed Opportunity

SPace Infrared telescope for Cosmology and Astrophysics



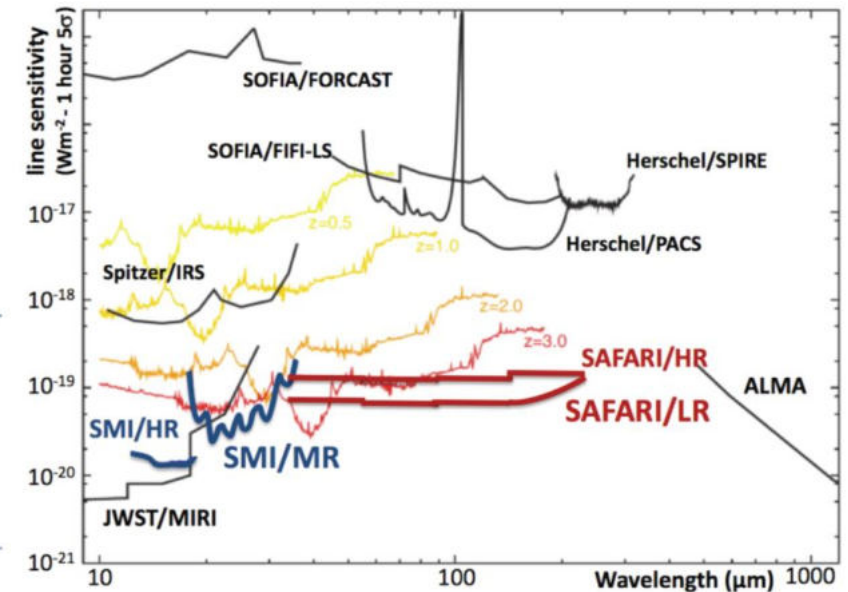
Launch: JAXA's H3 launch vehicle
Orbit: L2
Nominal mission lifetime: 3 years
Goal: 5 years



SAFARI: 34-230 μm background limited spectroscopy, $R\sim 300-11000$
100, 200, 350 μm imaging polarimeter

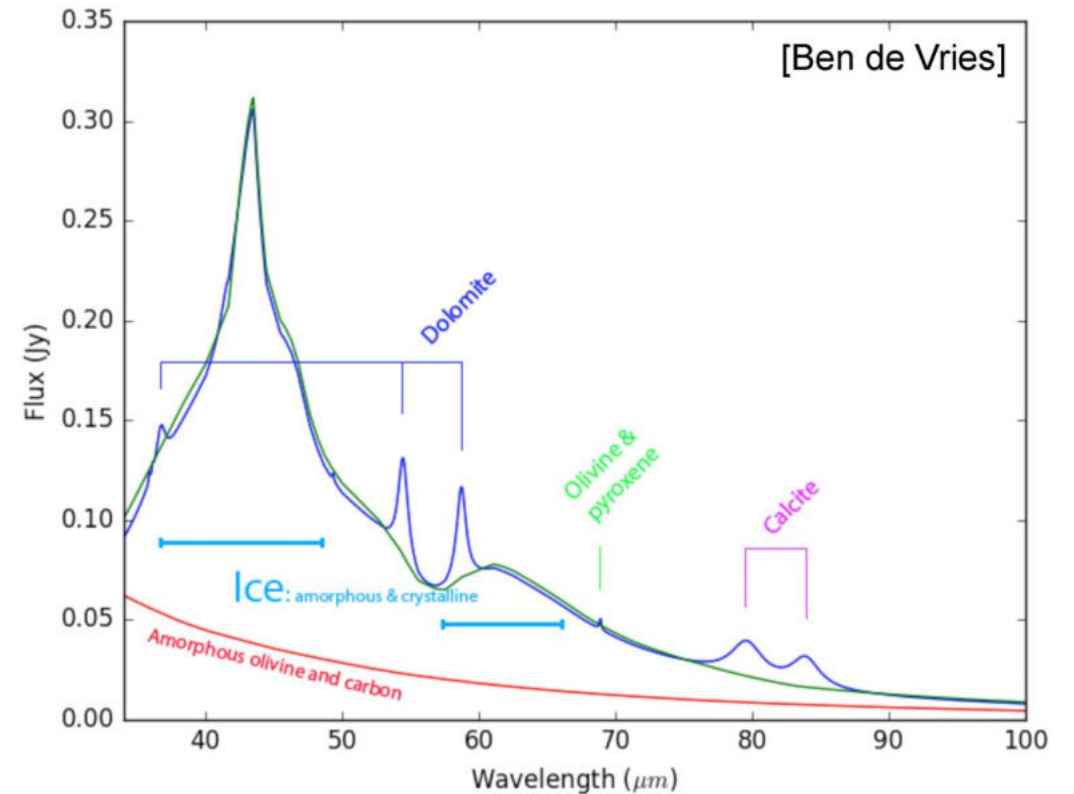
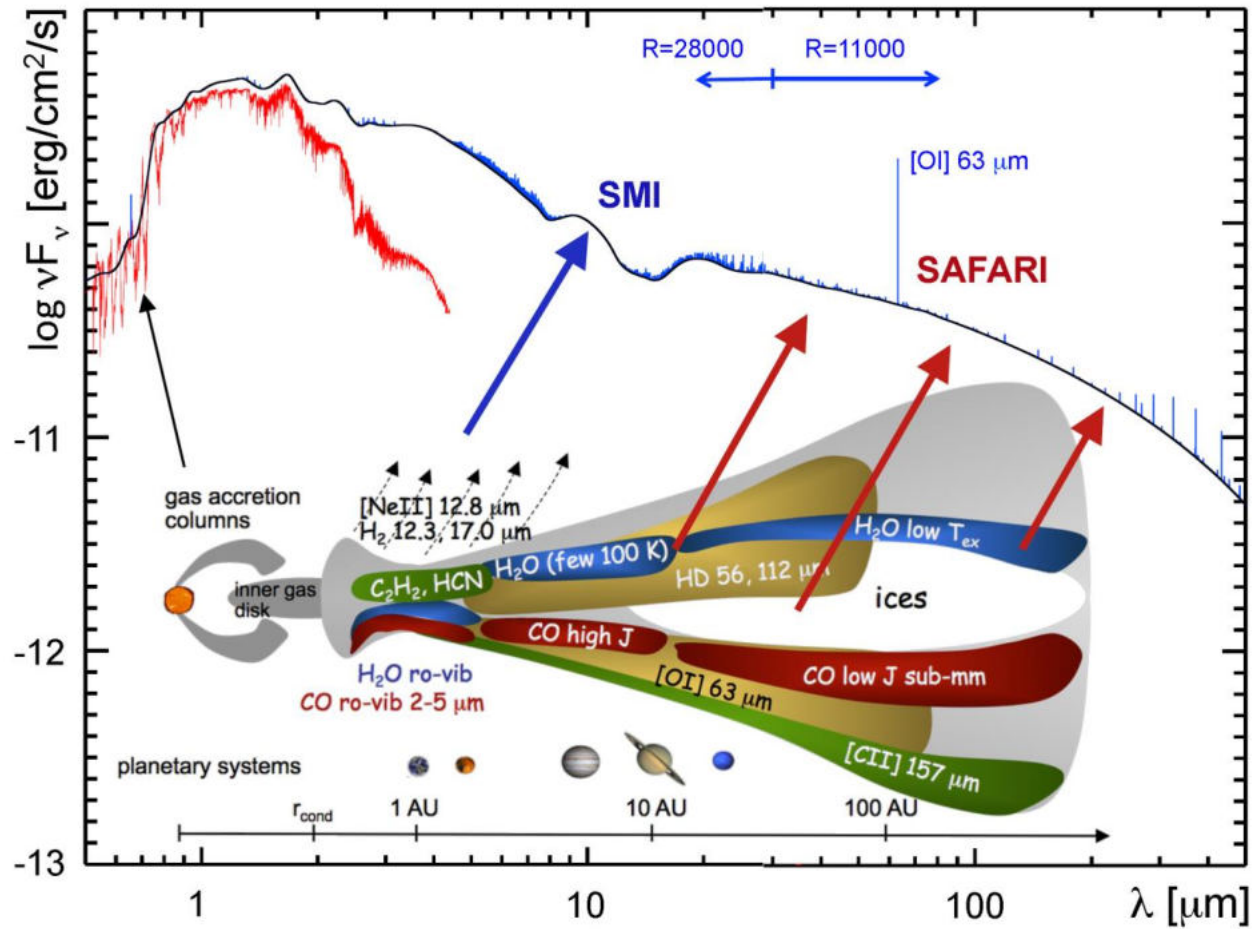
SMI: 17-36 μm background limited spectroscopy, $R\sim 100-2000$
12-18 μm spectroscopy at $R\sim 28000$
10'x12' survey camera and $R\sim 100$ spectrometer

2.5 meter <8K Ritchey-Chrétien telescope assembly
Mechanical coolers combined with passive V-groove cooling
JAXA H3 launcher



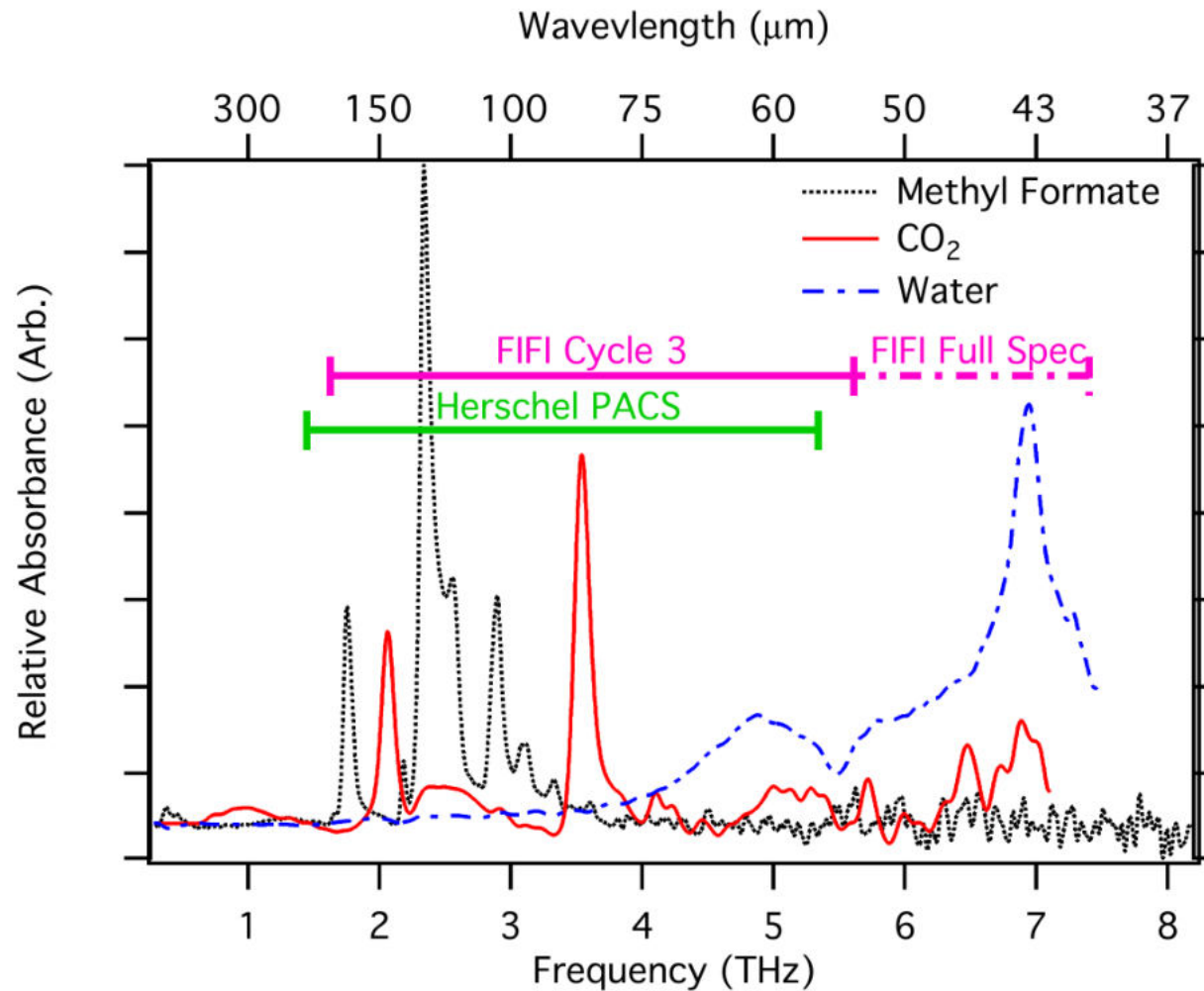
A Missed Opportunity

SPace Infrared telescope for Cosmology and Astrophysics

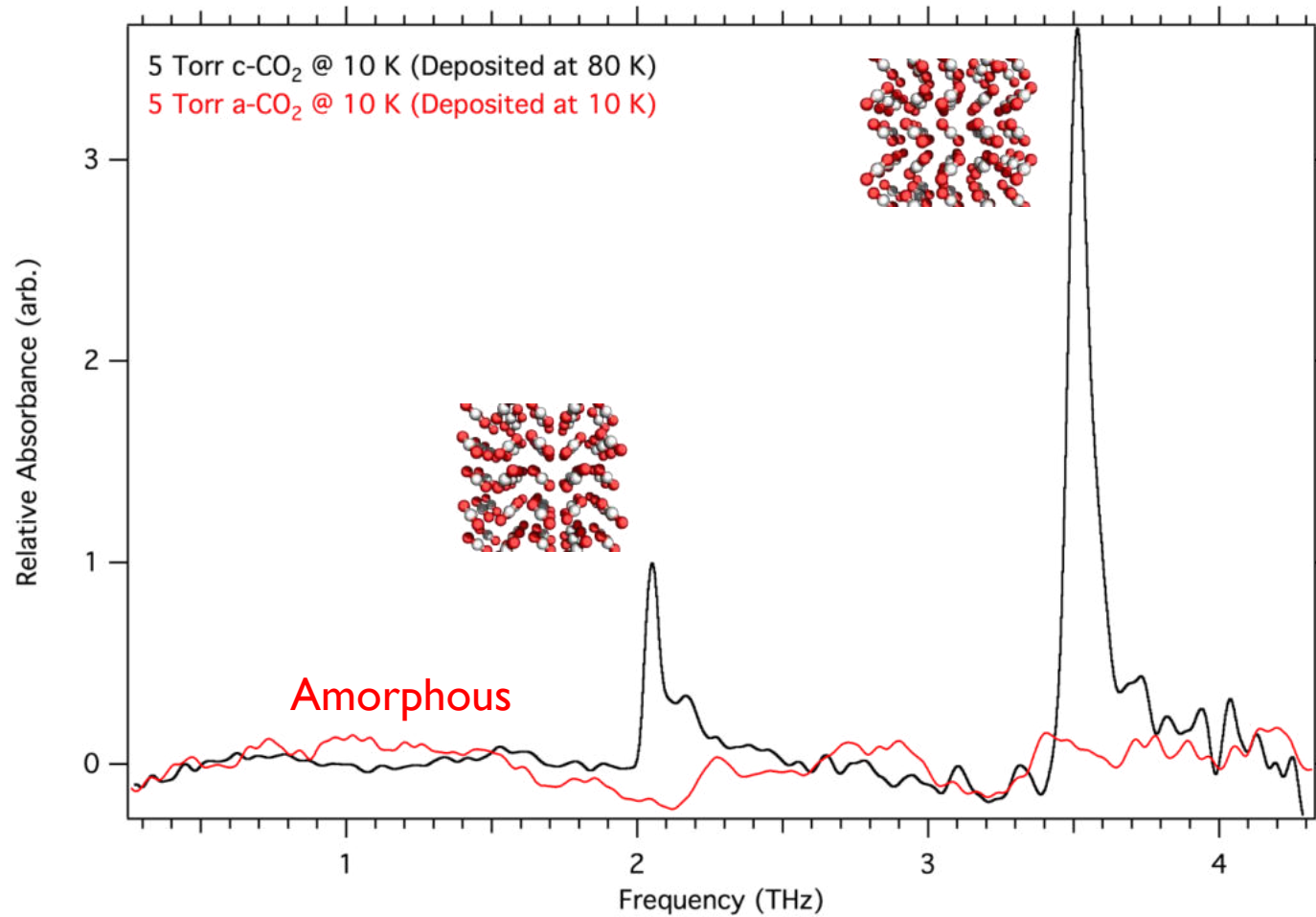


Potential Detection of COMs in the THz

28

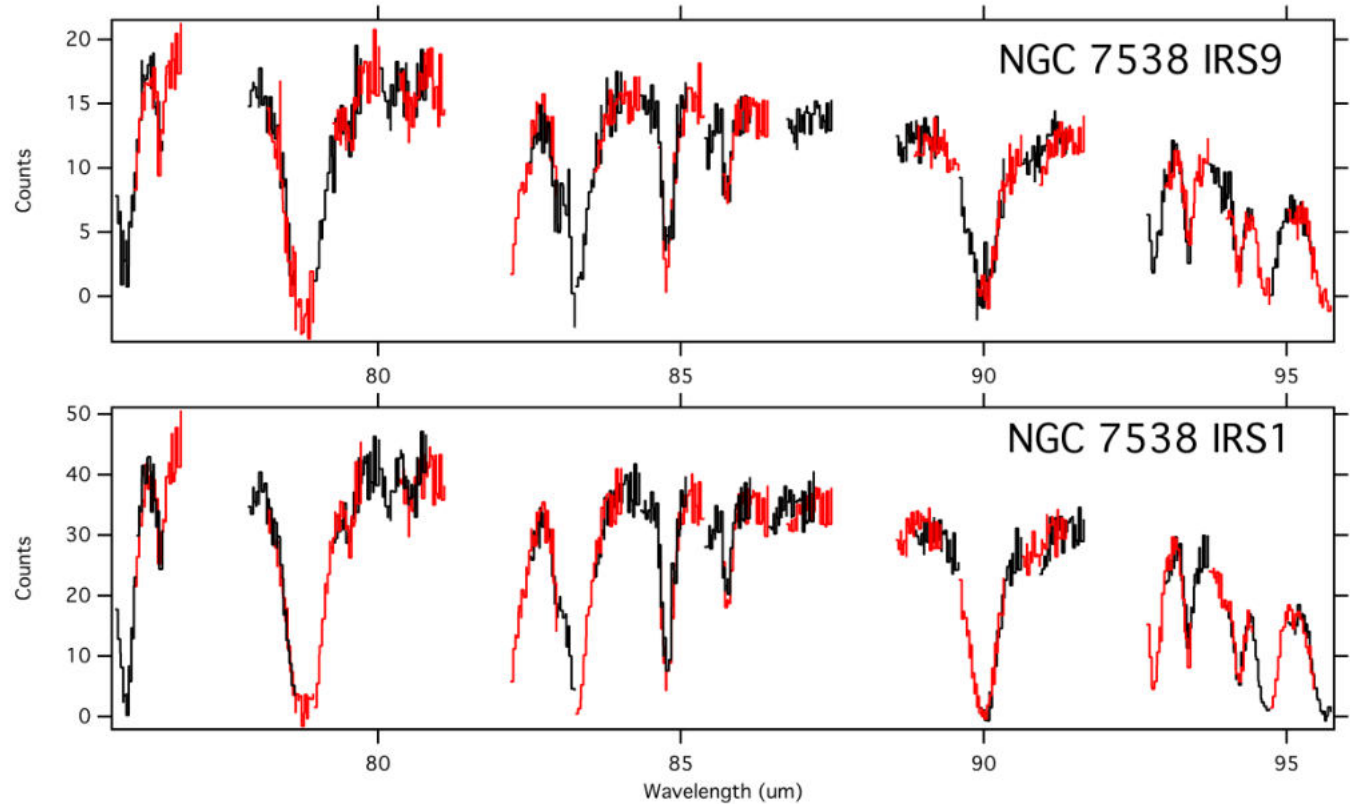


CO₂ - Amorphous vs crystalline Ice



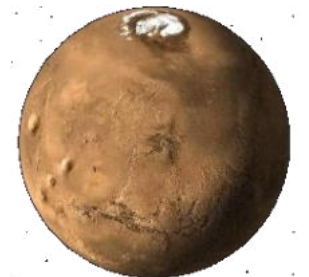
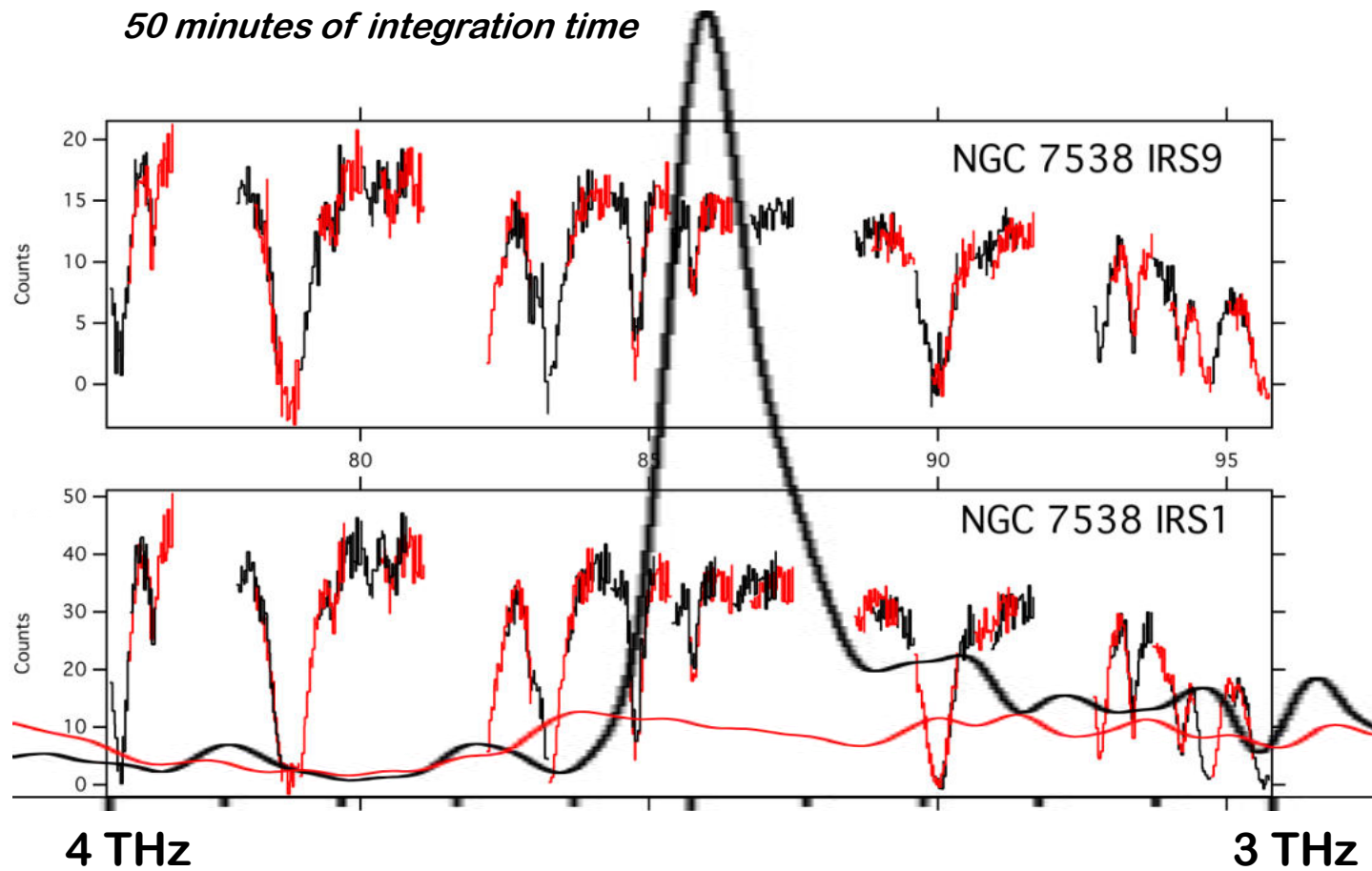
SOFIA Observations Cycle 3

50 minutes of integration time



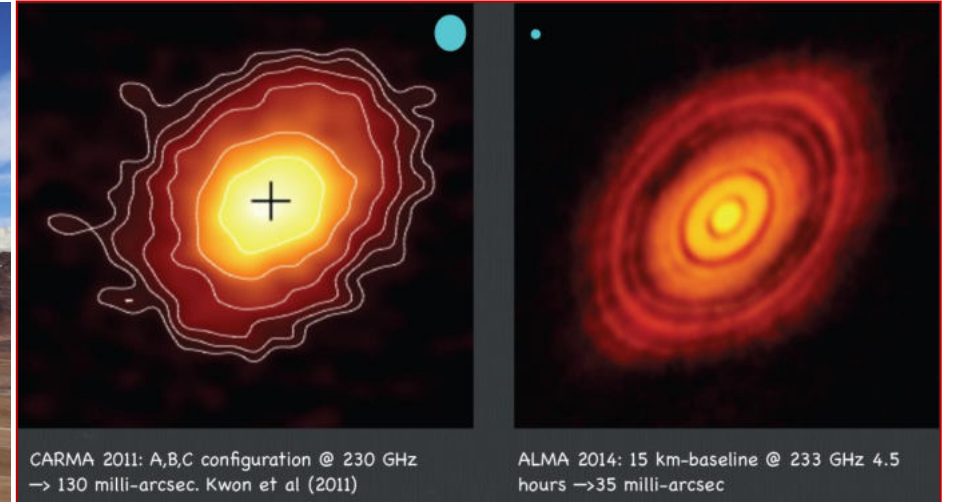
SOFIA Observations Cycle 3

50 minutes of integration time



ALMA - Atacama Large Millimeter/submillimeter Array

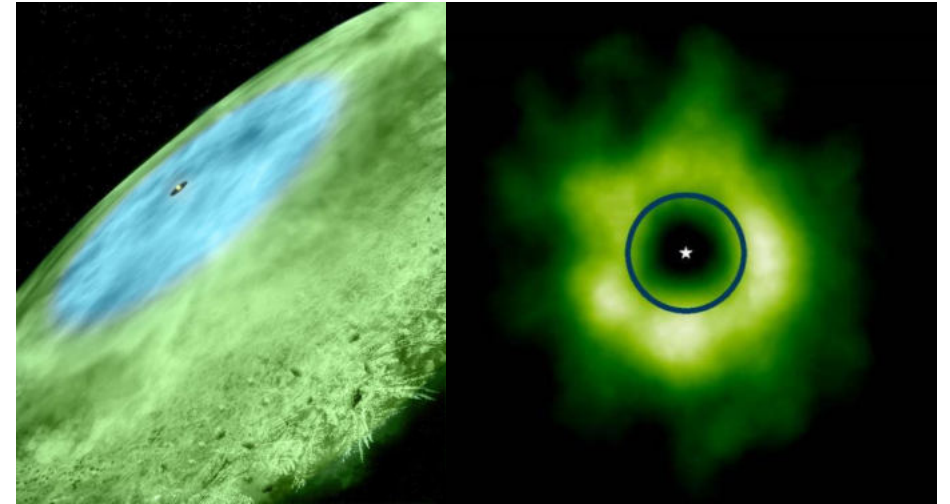
32



CREDIT: B. Saxton, NRAO/NSF

ALMA - Atacama Large Millimeter/submillimeter Array

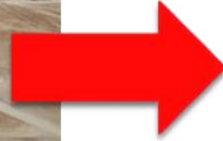
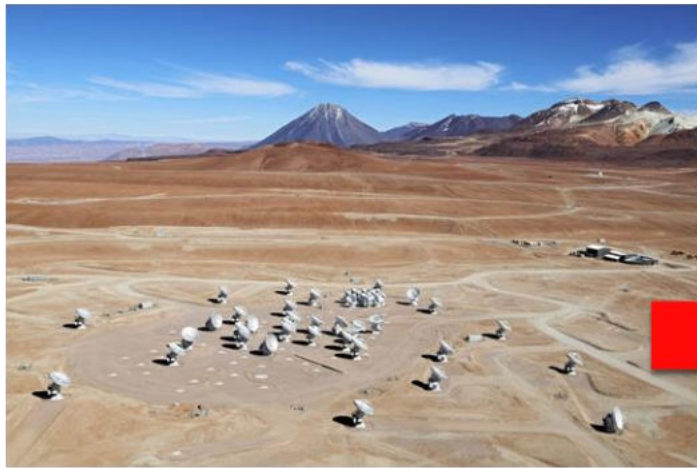
33



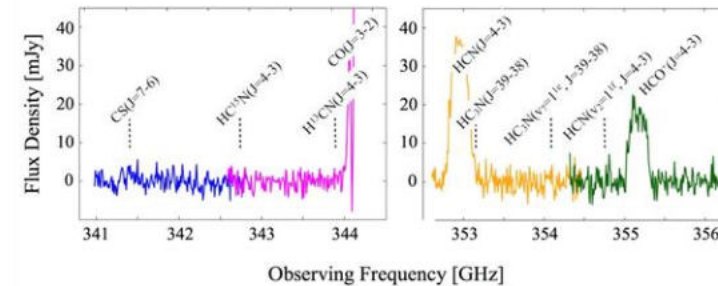
CREDIT: P. Salomé, Paris Obs.
K. Öberg, Harvard

ALMA - Atacama Large Millimeter/submillimeter Array Emission Spectroscopy

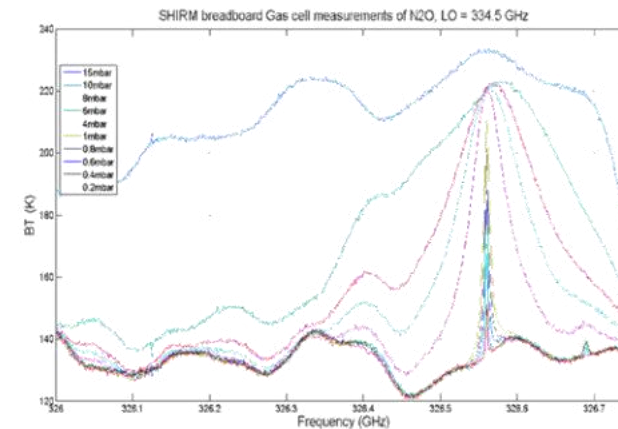
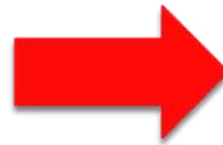
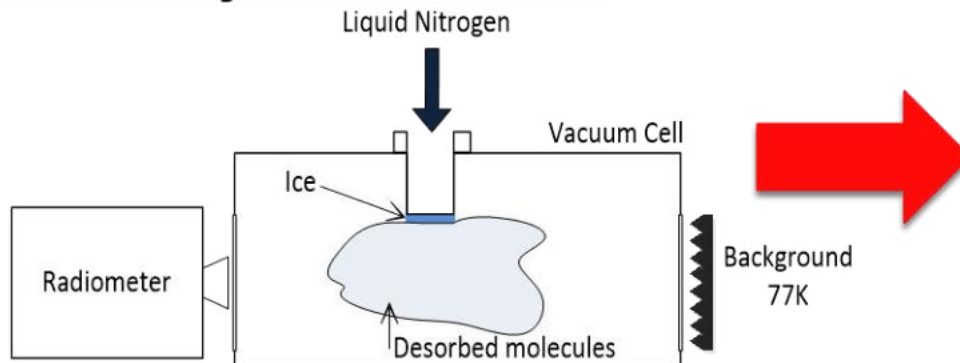
(sub)mm Observations



ALMA

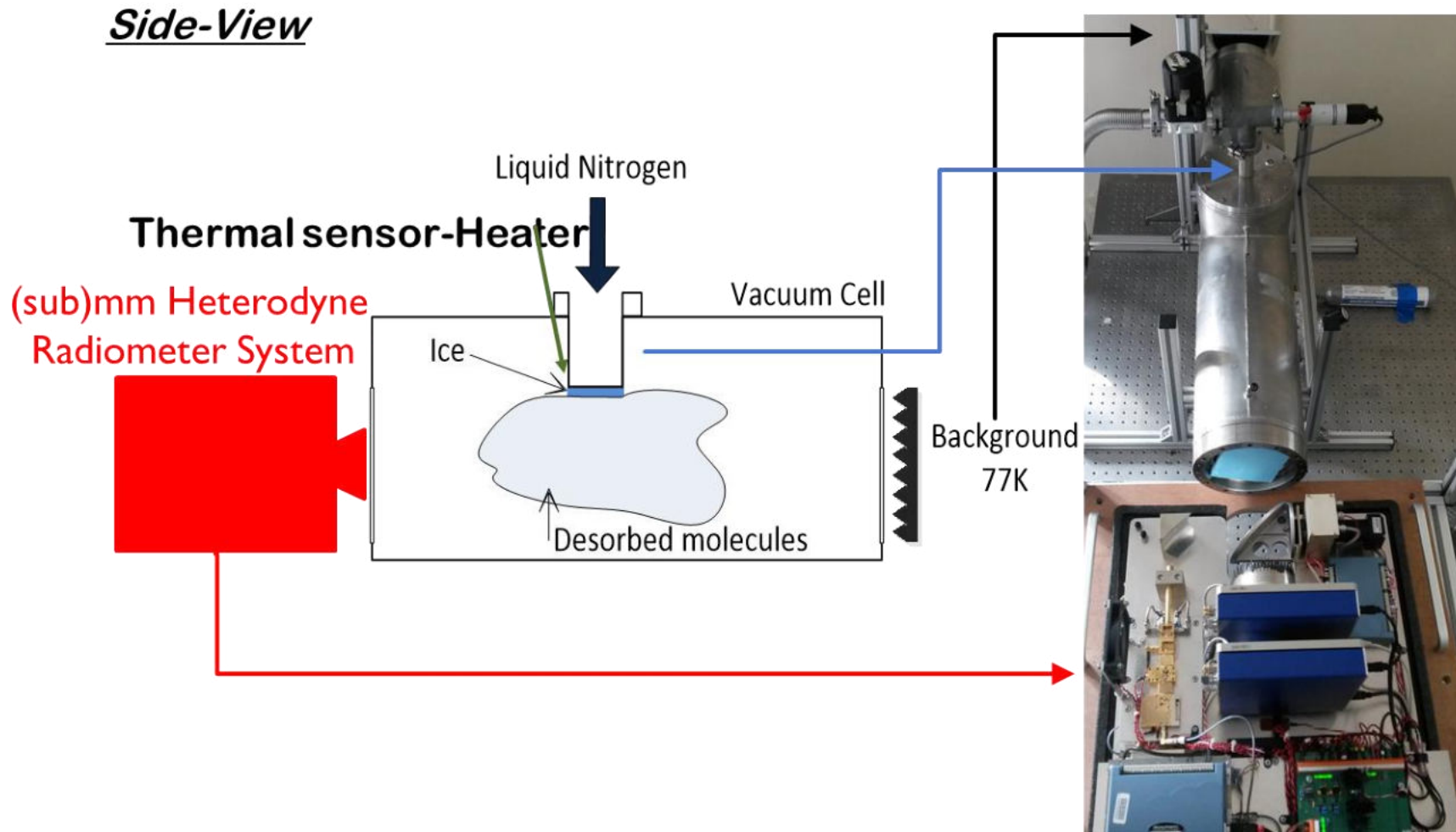


Laboratory Measurements

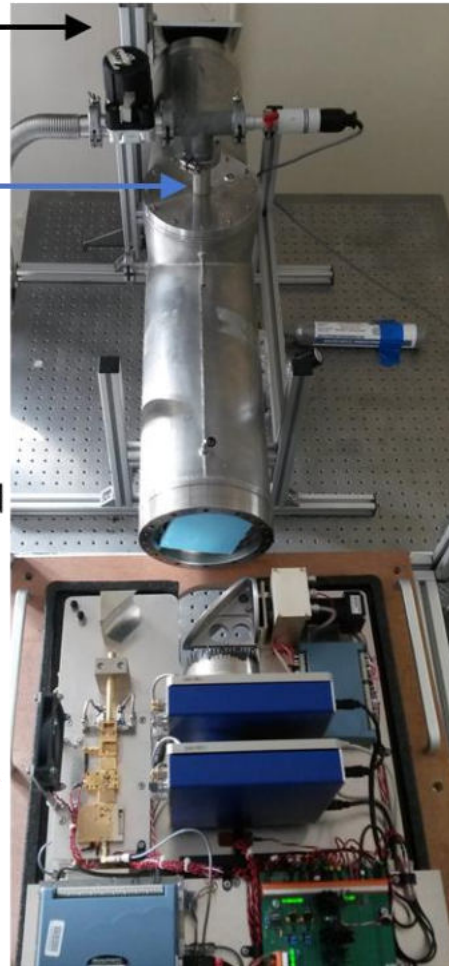


THz Desorption Emission Spectroscopy

35

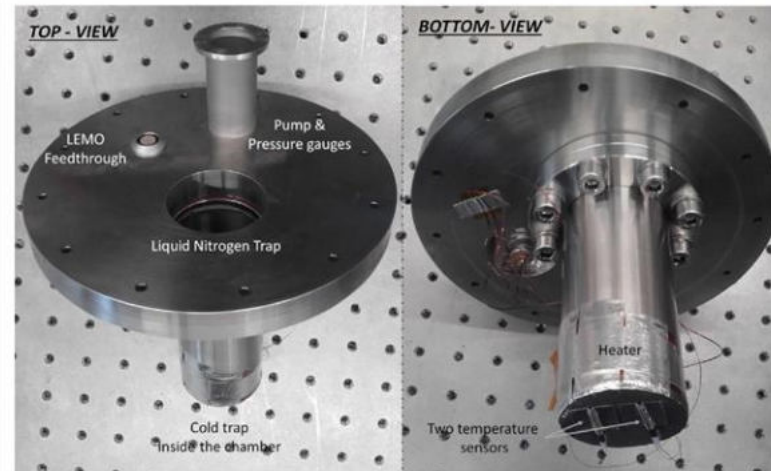
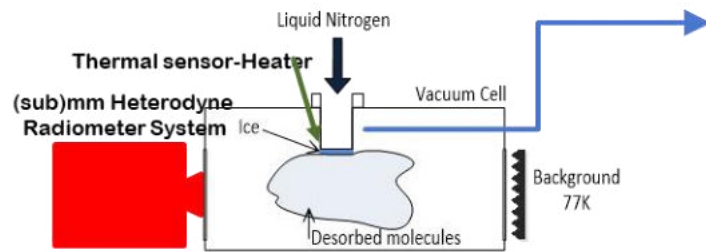


Top-View

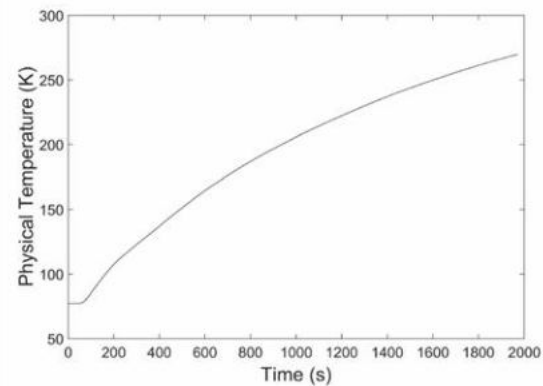


THz Desorption Emission Spectroscopy

Side-View

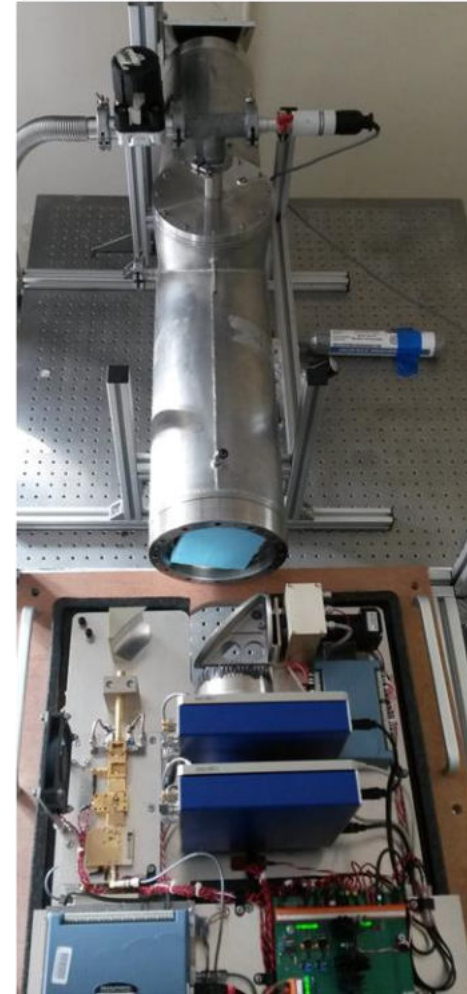
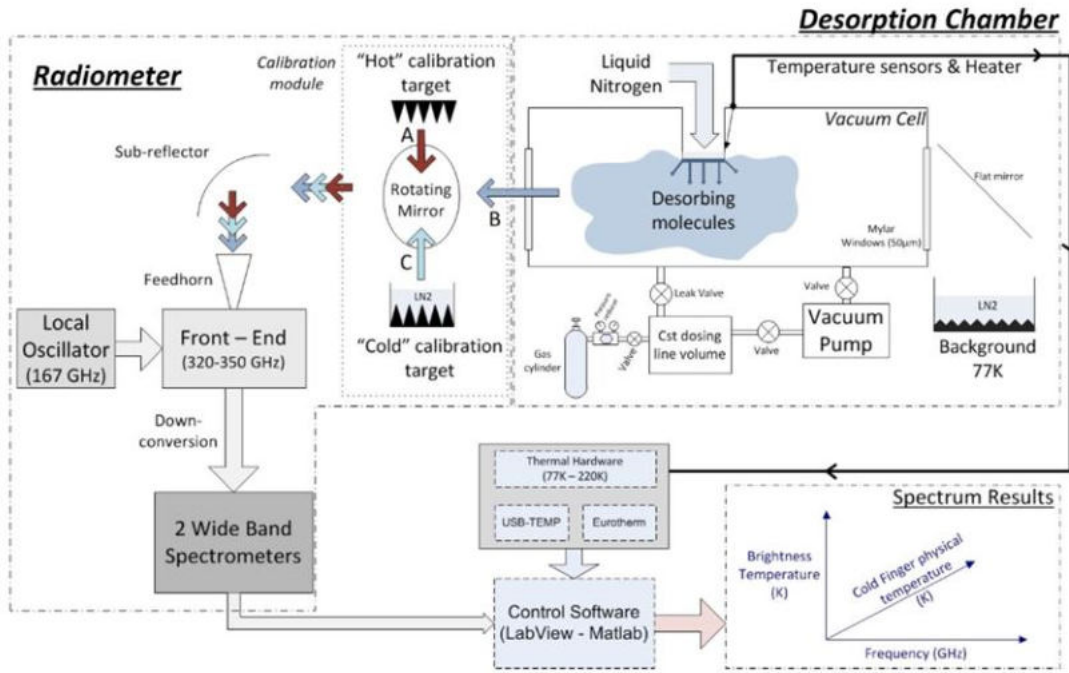


Auriacombe et al., MNRAS in prep.



THz Desorption Emission Spectroscopy

(sub)mm Heterodyne Radiometer System

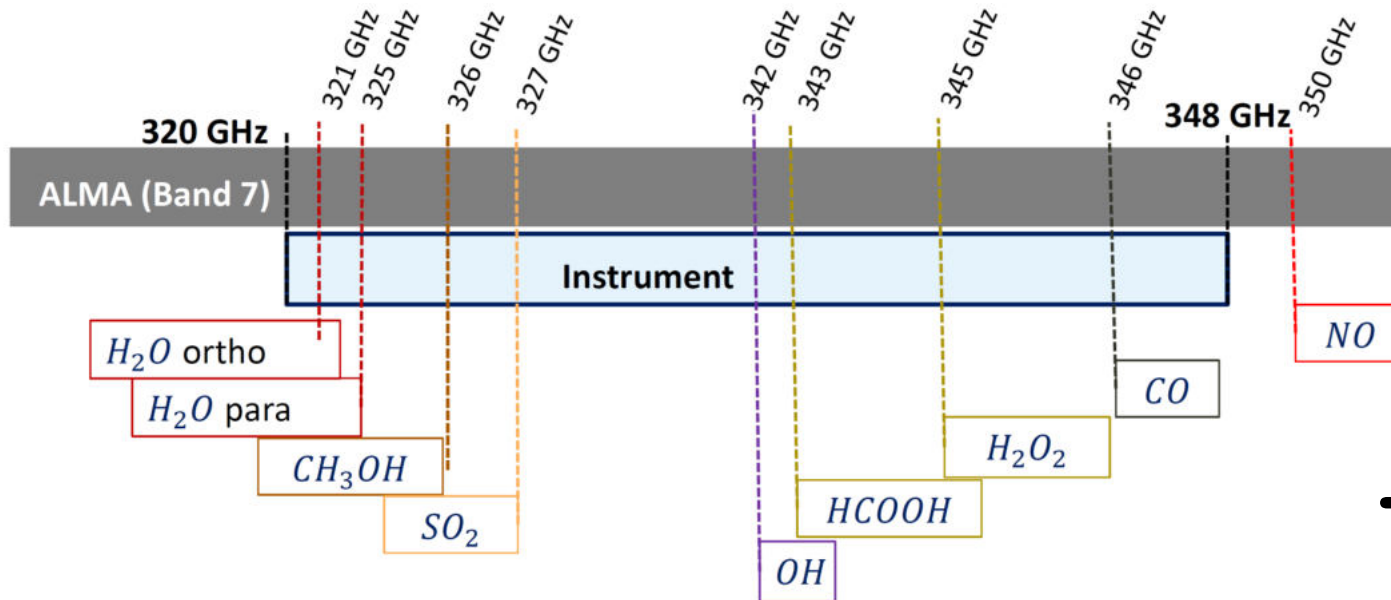


Top-View



THz Desorption Emission Spectroscopy

38



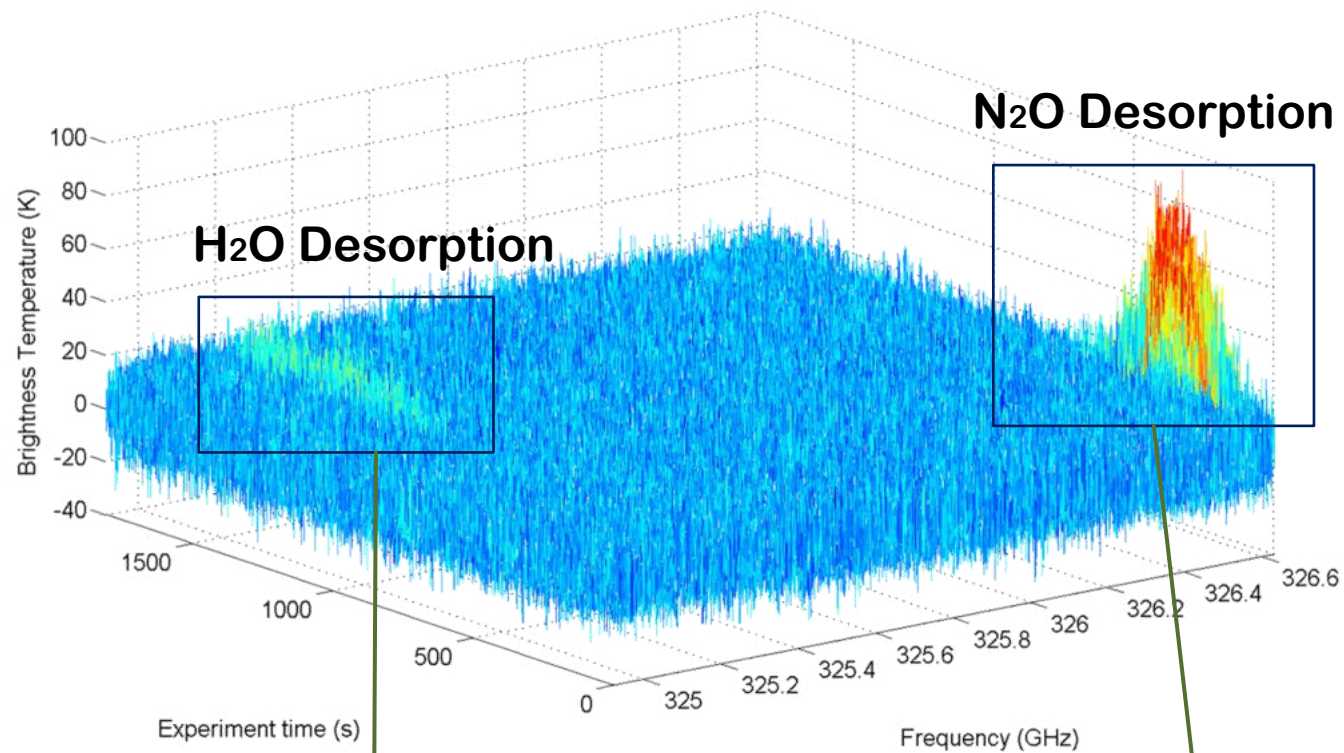
Target Molecules

Experiment:

Molecules	Frequency Line
Water	325,15 GHz
Methanol	326 GHz,...
Nitrous Oxide (N_2O)	326,55 GHz

THz Desorption Emission Spectroscopy

39



H₂O Desorption

- Start: ~ 230 K
- Peak: ~ 250 K

N₂O Desorption

- Start: ~ 90 K
- Peak: ~ 130 K

- **Need for airborne or spaceborne observatory instruments with appropriate bandwidth, resolution, and sensitivity.**
- **THz ice spectroscopy has the potential to identify the presence and temperature of COMs in interstellar ices and to characterize physics and chemistry of the ISM.**
- **THz Desorption Emission Spectroscopy can be used to study thermal desorption of COMs. More technology development needed to study non-thermal processes.**

Thanks for your attention!
Any questions?

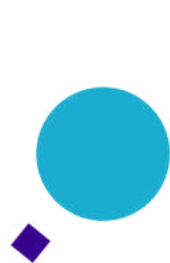
Key Papers:

- Allodi et al. (2014), *Phys. Chem. Chem. Phys.* **16**, 3442
- Ioppolo et al. (2016), *Faraday Discuss.* **168**, 461
- McGuire et al. (2016), *Phys. Chem. Chem. Phys.* **18**, 20199
- Mifsud et al. (2021), *Front. Astron. Space Sci.* **8**, 757619

University of
Kent


 Queen Mary
 University of London

Thanks to Collaborators at:



INAF
 ISTITUTO NAZIONALE DI ASTROFISICA
 OSSERVATORIO ASTROFISICO DI CATANIA



Radboud Universiteit



THE
**ROYAL
 SOCIETY**